NAMING ADDR

The word add evokes a sense of fear, stat in and of their. But, there should be no fear when naming an acid.

There are two types of acids we will be naming:

- Binary (meaning ago):
- 2. Oxyacid (meaning oxygen is present)

There are specific rules for both. The hardest part is identify what type of acid you have first. The easiest way is to first identify you have an acid (remember that ALL acids start with H in front, and the next is to look for what that H is bonding to, meaning the anion, or negative part of the acid.

BURNING ACTOR.

With a binary acid, the H will be bonded to a non-metal element, like chloring or bramine. This type of acid DOES NOT HAVE OXYGEN PRESENT IN THE FORMULA. A few of examples are listed below:

- · 194823
- A SHIPPING
- or Bellin

To name a binary acid there is a specific formula that is followed:

- Use the prefix hward e.
- Root of the anion's name +.
 - This extract
 - phosphorous: phosphor- is the root.
 - sulfur: sulfur- is the root
- Time suffix ic.

Lef's by to name some of those types of acids using this formula:

- w Example:
 - AND DESCRIPTION OF THE PERSON NAMED IN COLUMN TWO IN COLUM
 - hydro + ehlor (noot of chlorine) + & = hydrochigic acid

So, remember, binary acids have only two elements; hydrogen and a non-metal.

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A Loxley

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