

Covalent compounds- consist of nonmetal only.

If chemical formula starts with H- it's an acid: find its name or formula in the Acids List.
Exception: H_2O and H_2O_2 .

If chemical formula consist of 2 nonmetals only- use the following naming rules.

Ex: name the following compound - H_2O_2

- (a) name the first element- unchanged name from PT - hydrogen
 (b) for the second element, take **its root** and add ending **-ide** - ~~oxide~~
 (c) add prefixes which will show how many atoms of each element are here in a compound
 Answer: **dihydrogen dioxide**

Write the names for each of the following covalent compounds-

- | | |
|-------------------------------------|--|
| (1) NO - nitrogen monoxide | (2) HNO_3 - nitric acid |
| (3) H_3PO_4 _____ | (4) N_2O_5 _____ |
| (5) SiO_2 _____ | (6) N_2O_5 _____ |
| (7) HNO_2 _____ | (8) HBr _____ |
| (9) C_2H_6 _____ | (10) H_2SO_3 _____ |
| (11) CCl_4 _____ | (12) $\text{HC}_2\text{H}_3\text{O}_2$ _____ |
| (13) HClO_4 _____ | (14) HClO_3 _____ |
| (15) H_2SO_4 _____ | (16) H_2S _____ |
| (17) HI _____ | (18) H_2O_2 _____ |
| (19) Cl_3F _____ | (20) N_2H_4 _____ |
| (21) Cl_2O_7 _____ | (22) HCl _____ |
| (23) NH_3 _____ | (24) H_2SiO_4 _____ |
| (25) Cl_2O_3 _____ | (26) PCl_3 _____ |
| (27) HF _____ | (28) Cl_2O _____ |
| (29) CS_2 _____ | (30) H_2CO_3 _____ |
| (31) SF_6 _____ | (32) SiC _____ |
| (33) P_2S_5 _____ | (34) F_2O_3 _____ |

Naming Ionic Compounds Composition Answer Key

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