

Worksheet 5: Naming Ionic Compounds with Complex Ions

Name: _____

Please use your complex ion chart, along with your periodic table, to name these compounds.

Formula	Name of the Ionic Compound
1) AlCl_3	example: sodium hydroxide
2) $\text{Ba}(\text{NO}_3)_2$	
3) $(\text{NH}_4)_2\text{CO}_3$	
4) ZnSO_4	
5) Al_2O_3	
6) LiNO_3	
7) MgSO_4	
8) $\text{Sr}(\text{CN})_2$	
9) KClO_3	
10) Fe_2S_3	
11) $(\text{NH}_4)_3\text{P}$	
12) $\text{Ca}(\text{HCO}_3)_2$	
13) KBrO_4	
14) $\text{Zn}(\text{OH})_2$	
15) $\text{Hg}_2(\text{NO}_3)_2$	
16) AgNO_3	
17) $\text{Al}_2(\text{SO}_4)_3$	
18) BaCO_3	
19) $\text{Cr}(\text{ClO}_2)_2$	
20) Na_2PO_4	

Naming Compounds Containing Polyatomic Ions

Answers

Morris Hein, Susan Arena, Cary Willard



Naming Compounds Containing Polyatomic Ions Answers:

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Regents Chemistry--Physical Setting Power Pack Revised Edition Barron's Educational Series, Albert S. Tarendash, 2021-01-05 Barron's two book Regents Chemistry Power Pack provides comprehensive review actual administered exams and practice questions to help students prepare for the Chemistry Regents exam This edition includes Regents Exams and Answers Chemistry Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day Let's Review Regents Chemistry Extensive review of all topics on the test Extra practice questions with answers A detailed introduction to the Regents Chemistry course and exam One actual recently released Regents Chemistry exam with an answer key

Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice

resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields

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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone s Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into

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undergraduate students and advanced high school students of chemistry
Smoot, Smoot, Richard G. Smith, Jack Price, 1998

Merrill Chemistry Robert C.

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Table of Contents Naming Compounds Containing Polyatomic Ions Answers

1. Understanding the eBook Naming Compounds Containing Polyatomic Ions Answers
 - The Rise of Digital Reading Naming Compounds Containing Polyatomic Ions Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Naming Compounds Containing Polyatomic Ions Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Naming Compounds Containing Polyatomic Ions Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Naming Compounds Containing Polyatomic Ions Answers
 - Personalized Recommendations
 - Naming Compounds Containing Polyatomic Ions Answers User Reviews and Ratings
 - Naming Compounds Containing Polyatomic Ions Answers and Bestseller Lists
5. Accessing Naming Compounds Containing Polyatomic Ions Answers Free and Paid eBooks
 - Naming Compounds Containing Polyatomic Ions Answers Public Domain eBooks
 - Naming Compounds Containing Polyatomic Ions Answers eBook Subscription Services
 - Naming Compounds Containing Polyatomic Ions Answers Budget-Friendly Options
6. Navigating Naming Compounds Containing Polyatomic Ions Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Naming Compounds Containing Polyatomic Ions Answers Compatibility with Devices
 - Naming Compounds Containing Polyatomic Ions Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Naming Compounds Containing Polyatomic Ions Answers
 - Highlighting and Note-Taking Naming Compounds Containing Polyatomic Ions Answers
 - Interactive Elements Naming Compounds Containing Polyatomic Ions Answers
 8. Staying Engaged with Naming Compounds Containing Polyatomic Ions Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Naming Compounds Containing Polyatomic Ions Answers
 9. Balancing eBooks and Physical Books Naming Compounds Containing Polyatomic Ions Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Naming Compounds Containing Polyatomic Ions Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Naming Compounds Containing Polyatomic Ions Answers
 - Setting Reading Goals Naming Compounds Containing Polyatomic Ions Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Naming Compounds Containing Polyatomic Ions Answers
 - Fact-Checking eBook Content of Naming Compounds Containing Polyatomic Ions Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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