

2015

Subject : Chemistry (Honours)

Paper : V

Time: 2 Hours

Full Marks: 50

*The figures in the margin indicate full marks.
Candidates are required to give their answers in their own words
as far as practicable.*

1. Answer any four questions:

2×4=8

- Depict the ambidentate and flexidentate behaviours of thiocyanate anion.
- SiCl_4 has lower boiling point than CCl_4 . Explain.
- Cite and draw one example each of $3c-2e$ and $2c-2e$ bonds.
- Briefly comment on the compound K_3C_{60} .
- Write down the structural formulae of sodium pyrophosphate and sodium metaphosphate.
- What is meant by chalcogenide? Give example.
- 'NaOH is base but ClOH is acid' - Give reason.

2. Answer any three questions:

6×3=18

- What are inner metallic complexes? Give example. 2
 - Chromium(III) chloride forms three different hydrates of the same mole ratio $\text{Cr:Cl:H}_2\text{O}=1:3:6$. The violet form does not lose water over conc. H_2SO_4 and gives 3 equivalent of AgCl on treatment with AgNO_3 . The other two forms are green and lose 1 and 2 moles of H_2O over conc. H_2SO_4 and give 2 and 1 equivalent of AgCl , respectively with AgNO_3 . Write down the structures of the three isomeric coordination compounds and explain the isomerism involved. 4
- Define and exemplify chelate complex. 3
 - 'The stability of $[\text{Ni}(\text{en})_3]^{2+}$ (en =ethylenediamine) is much greater than that of $[\text{Ni}(\text{NH}_3)_6]^{2+}$, although both contain six Ni-N bonds.' - Explain. 3
- Apply radius ratio principle to explain the difference between crystal structures of NaCl and CsCl . 2
 - ' MgO is a stable ionic solid though large amount of energy is required to form Mg to Mg^{2+} and to dissociate O_2 into atomic oxygen and thereafter to form O^{2-} ion.' - Account for the fact. 2

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Raffaella Di Napoli



Microbiology University Of Burdwan Previous Year Question Papers:

The Top Books of the Year Microbiology University Of Burdwan Previous Year Question Papers The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous engrossing novels captivating the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the captivating narratives that have enthralled audiences this year.

Microbiology University Of Burdwan Previous Year Question Papers : Colleen Hoover's "It Ends with Us" This touching tale of love, loss, and resilience has gripped readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed.

Uncover the Best : Taylor Jenkins Reid's "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reid's compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery.

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These popular novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered.

The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts.

The Secret History is an exceptional and suspenseful novel that will keep you speculating until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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