

# AQA Chemistry

## GCSE Exam-style mark scheme

## C4 Chemical calculations

100 marks (20%)

1 hour

Question	Answer	Mark	Mark	Mark
10.1.1	14.2 g		1	100
10.1.2	14.2 g	<p>Amount of magnesium carbonate  <math>14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}</math>  <math>= 0.1 \text{ mol}</math>  <math>\therefore 0.1 \text{ mol} \times 142.2 \text{ g mol}^{-1}</math>  <math>= 14.22 \text{ g}</math>  <math>\therefore 14.2 \text{ g}</math></p>	2	100
10.1.3	<p>any 1000 factor</p> <ul style="list-style-type: none"> <li>converting grams</li> <li>into kg (multiplication)</li> <li>needed</li> <li>into kg (multiplication) needed</li> <li>into the equation</li> </ul>		2	100
10.2	<p>all calculations must be            shown            answers all need to have the            units (multiplication)            all will endowentire (units)            into kg (multiplication) to            make (units) added            together</p>	<p>Amount of magnesium carbonate  <math>14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}</math>  <math>= 0.1 \text{ mol}</math>  <math>\therefore 0.1 \text{ mol} \times 142.2 \text{ g mol}^{-1}</math>  <math>= 14.22 \text{ g}</math>  <math>\therefore 14.2 \text{ g}</math></p>	2	100
10.3.1	<p>all calculations must be            shown</p>		2	100
10.3.2	<p>all calculations must be            shown</p>		2	100
10.3.3	<p>all calculations must be            shown</p>		2	100

# Jan 2014 C4 Mark Scheme International Paper

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