## HSC Maharashtra Board question paper: March 2013

#### Note:

- All question are compulsory
- ii. Answer to the two sections are to be written in the same answer book.
- iii. Figure to the right hand side indicate full marks.
- Write balanced chemical equations and draw neat and labelled diagrams wherever necessary.
- Encery new question must be started on a new page.
- vi. Use of logarithmic table is allowed

# CHEMISTRY: SECTION - I

#### Q. 1. Select and write the most appropriate answer from the given alternatives for each sub-question: [7

i. Ir	n body centred cubic structure the space occupied is about				
(2	(A)	68 %		(B)	53 %
(0	C).	38 %		(D)	32 %
ii. Fe	For a gaseous reaction the unit of rate of reaction is				
(2	A)	L atm s <sup>-1</sup>		(B)	arm mol <sup>-1</sup> s <sup>-1</sup>
(6	C)	atm s 1		(D)	mol s
iii. W	Which of the following compounds contains S = O as well as S = S bonds?				
C2	(A)	Sulphuric acid		(B)	Thiosulphuric acid
(6	C)	Sulphurous acid	00	(D)	Thiosulphurous acid
iv. W	Which of the following solutions shows maximum depression in freezing point?				
(4	(A)	0.5 M Li <sub>2</sub> SO <sub>4</sub>		(B)	1 M NaCl
0	C)	0.5 M Al <sub>2</sub> (SO <sub>4</sub> ) <sub>5</sub>	) "	(D)	0.5 M BaCl <sub>2</sub>
v. F	For a chemical reaction $\Delta S = -0.035$ kJ/K and $\Delta H = -20$ kJ.				
(4	A)	5.14 K		(B)	57.14 K
6	2)	571.4 K		(D)	5714.0 K
vi. T	The standard e.m.f of the following cell is 0.463 V				
C	Su	Cum   Agan   Ag. I	If Eas - 0.800 V,		

vii. Fe<sub>2</sub>O<sub>3</sub> is reduced to spongy iron near the top of blast furance by

(A) H<sub>2</sub>

(A) 1.137 V

(C) 0.463 V

(B) CaO

0.337 V

-0.463 V

(B)

(ID)

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What is the standard potential of Cu electrode?

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#### **Table of Contents Hsc Board Chemistry 1 Question Paper 2013**

- 1. Understanding the eBook Hsc Board Chemistry 1 Question Paper 2013
  - The Rise of Digital Reading Hsc Board Chemistry 1 Question Paper 2013
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Hsc Board Chemistry 1 Question Paper 2013
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Hsc Board Chemistry 1 Question Paper 2013
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Hsc Board Chemistry 1 Question Paper 2013
  - Personalized Recommendations
  - $\circ\,$  Hsc Board Chemistry 1 Question Paper 2013 User Reviews and Ratings
  - Hsc Board Chemistry 1 Question Paper 2013 and Bestseller Lists
- 5. Accessing Hsc Board Chemistry 1 Question Paper 2013 Free and Paid eBooks
  - Hsc Board Chemistry 1 Question Paper 2013 Public Domain eBooks
  - Hsc Board Chemistry 1 Question Paper 2013 eBook Subscription Services
  - Hsc Board Chemistry 1 Question Paper 2013 Budget-Friendly Options

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  - o ePub, PDF, MOBI, and More
  - Hsc Board Chemistry 1 Question Paper 2013 Compatibility with Devices
  - Hsc Board Chemistry 1 Question Paper 2013 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Hsc Board Chemistry 1 Question Paper 2013
  - Highlighting and Note-Taking Hsc Board Chemistry 1 Question Paper 2013
  - Interactive Elements Hsc Board Chemistry 1 Question Paper 2013
- 8. Staying Engaged with Hsc Board Chemistry 1 Question Paper 2013
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Hsc Board Chemistry 1 Question Paper 2013
- 9. Balancing eBooks and Physical Books Hsc Board Chemistry 1 Question Paper 2013
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Hsc Board Chemistry 1 Question Paper 2013
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Hsc Board Chemistry 1 Question Paper 2013
  - Setting Reading Goals Hsc Board Chemistry 1 Question Paper 2013
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Hsc Board Chemistry 1 Question Paper 2013
  - Fact-Checking eBook Content of Hsc Board Chemistry 1 Question Paper 2013
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

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