

## HONORS CHEM 1st SEMESTER EXAM STUDY GUIDE FALL 2010

**STRATEGY:** Start by reading through your notes to refresh your memory on these topics. Then, use this review sheet as a starting point to identify the areas on which you need to spend more study time. For those areas, go back to homework assignments, quizzes, and reviews to practice more problems. I would also recommend going through all of your tests since these questions are only samples and do not include specific examples of how vocabulary and other conceptual information might appear in a multiple-choice or other format. Remember you can access notes and reviews on my website.

**IMPORTANT INFORMATION:**

- This must be handed in BEFORE taking your final exam.
- If entirely complete, 5 points will be added to your exam grade
- All work must be shown in DETAIL on lined paper

**Introduction to Chemistry – Chapter 1**

1. Identify safe/unsafe lab practices.
2. List pieces and uses of lab equipment.
3. List the requirements for tables and graphs.
4. What is plotted on the x-axis and on the y-axis?
5. Make a density graph.
6. What is water displacement?

**Matter and Change – Chapter 2**

7. Classify the following substances as *solid*, *liquid*, *gas*, or *plasma* based on their properties.
  - a. flexible volume, high KE, particles can disperse freely.
  - b. flexible volume, very high KE, particles are charged.
  - c. fixed volume, very low KE, orderly particles.
  - d. fixed volume, low KE, particles can move past each other.
8. Classify the following as element, compound, heterogeneous mixture, or solution.
  - a. graphite (carbon)
  - b. grape juice
  - c. table salt (NaCl)
  - d. pepper
9. Classify the following as *chemical* or *physical changes*.
  - a. cutting wire
  - b. ripening tomato
  - c. apple slices turning brown
  - d. compressing a gas
10. Classify the following properties as *physical* or *chemical*.
  - a. melts at 68.0°C
  - b. corrosive
  - c. reacts violently with water
  - d. decomposes in air
  - e. magnetic
11. Classify the following properties as *intensive* or *extensive*.
  - a. melting point
  - b. volume
  - c. mass
  - d. reactivity
  - e. density

**VOCAB:** kinetic molecular theory

# Honors Chemistry Final Exam Study Guide

**S Baum**



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## **Table of Contents Honors Chemistry Final Exam Study Guide**

1. Understanding the eBook Honors Chemistry Final Exam Study Guide
  - The Rise of Digital Reading Honors Chemistry Final Exam Study Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying Honors Chemistry Final Exam Study Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Honors Chemistry Final Exam Study Guide
  - User-Friendly Interface
4. Exploring eBook Recommendations from Honors Chemistry Final Exam Study Guide
  - Personalized Recommendations
  - Honors Chemistry Final Exam Study Guide User Reviews and Ratings
  - Honors Chemistry Final Exam Study Guide and Bestseller Lists
5. Accessing Honors Chemistry Final Exam Study Guide Free and Paid eBooks
  - Honors Chemistry Final Exam Study Guide Public Domain eBooks
  - Honors Chemistry Final Exam Study Guide eBook Subscription Services
  - Honors Chemistry Final Exam Study Guide Budget-Friendly Options

6. Navigating Honors Chemistry Final Exam Study Guide eBook Formats
  - ePub, PDF, MOBI, and More
  - Honors Chemistry Final Exam Study Guide Compatibility with Devices
  - Honors Chemistry Final Exam Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Honors Chemistry Final Exam Study Guide
  - Highlighting and Note-Taking Honors Chemistry Final Exam Study Guide
  - Interactive Elements Honors Chemistry Final Exam Study Guide
8. Staying Engaged with Honors Chemistry Final Exam Study Guide
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Honors Chemistry Final Exam Study Guide
9. Balancing eBooks and Physical Books Honors Chemistry Final Exam Study Guide
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Honors Chemistry Final Exam Study Guide
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Honors Chemistry Final Exam Study Guide
  - Setting Reading Goals Honors Chemistry Final Exam Study Guide
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Honors Chemistry Final Exam Study Guide
  - Fact-Checking eBook Content of Honors Chemistry Final Exam Study Guide
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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