

A) Describe the Kinetic Molecular Theory of Gas

Kinetic Molecular Theory

- tiny particles, atoms, or molecules moving at high velocities
- most volume of gas is empty space
- no attractive or repulsive forces between gas particles
- gas particles are constantly moving in straight paths; collisions occur with complete transfer of energy
- the average kinetic energy of gas particles is proportional to the Kelvin temp of sample

B) Identify and convert between the units of measurements used for gases

* on equation sheet

C) Use the ideal gas law to solve for P, V, T, or n of a gas

Ideal gas law: $PV = nRT$ P = pressure
V = volumeT = temperature
n = amount
(number of gas)
in molesR = ideal gas constant
= $0.0821 \frac{\text{L} \cdot \text{atm}}{\text{K} \cdot \text{mol}}$ * given
on equation
sheet

D) Use the combined gas law

$$\frac{P_1 \times V_1}{n_1 \times T_1} = \frac{P_2 \times V_2}{n_2 \times T_2}$$

E) Recognize

Inverse RelationshipP and V
T and P

$$\frac{P \times V}{n \times T}$$

Direct RelationshipV and T
n and V

$$\frac{P \times V}{n \times T}$$

F) STP

Standard Temperature + Pressure

↓
0°C
273.15 K

↓
1 atm

G) Use the molar volume at STP as a conversion factor

$$\star \frac{1 \text{ mol gas}}{22.4 \text{ L}} \quad \text{or} \quad \frac{22.4 \text{ L}}{1 \text{ mol gas}}$$

Guided Practice Problems Chapter 8 Chemistry

Craig T. Smith



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Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples, and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step by step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid base reactions, which often persists through working with acid base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshmen in STEM fields.

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes. Also includes solutions and explanations to odd numbered exercises.

Survival Guide to Organic Chemistry Patrick E. McMahon, Bohdan B. Khomtchouk, Claes Wahlestedt, 2016-12-19 The Survival Guide to Organic Chemistry: Bridging the Gap from General Chemistry enables organic chemistry students to bridge the gap between general chemistry and organic chemistry. It makes sense of the myriad of in-depth concepts of organic chemistry without overwhelming them in the necessary detail often given in a complete organic chemistry text. Here the topics covered span the entire standard organic chemistry curriculum. The authors describe subjects which require further explanation, offer alternate viewpoints for understanding, and provide hands-on practical problems and solutions to help master the material. This text ultimately allows students to apply key ideas from their general chemistry curriculum to key concepts in organic chemistry. Key Features: Reviews key general chemistry concepts and techniques adapted for application to important organic principles. Provides practical guidance to help students make the notoriously well-known and arduous transition from general chemistry to organic chemistry. Explains organic concepts and reaction mechanisms, generally expanding the focus on how to understand each step from a more intuitive viewpoint. Covers concepts that need further explanation as well as those that summarize and emphasize key ideas or skills necessary in this field. An added bonus is help with organizing principles to make sense of a wide range of similar reactions and mechanisms. Implements a user-friendly process to achieve the end result of problem solving. Covers organic chemistry I and II concepts at the level and depth of a standard ACS organic chemistry curriculum.

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