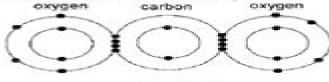


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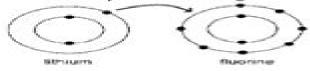
- I. MULTIPLE CHOICE INSTRUCTION: Encircle the letter of the correct answer. (1 POINT EACH)
- 1. It is a subatomic particle, symbol e- or
 - β-, with a negative elementary electric charge.
 - A. Neutron
 - 8. Electron
 - C. Protom
 - D. Attorni
- It is a subatomic particle, symbol p or p+, with a
 positive electric charge of +1e elementary
 charge and mass slightly less than that of a
 meutron.
 - A. Neutron
 - 8. Electron
 - C. Protom
 - D. Altern
- It is a subatomic particle, symbol n or nO, with no net electric charge and a mass slightly larger than that of a proton.
 - A. Neutron B. Electron
 - C. Proton
 - D. Atom
- 4. Is this an example of ionic bonding?



carbon dioxide

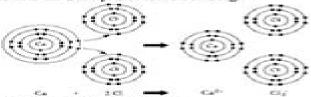
- A. Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- C. There is no chemical bonding present.
- D. This is just an orbital model.

5. Is this an example of ionic bonding?



littreum fluoride

- A. Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- D. This is just an orbital model.
- In atomic physics and quantum chemistry, the electron configuration is the distribution of electrons of an atom or molecule in atomic or molecular orbitals.
 - A. Electron Configuration
 - Periodic Table of Elements
 - C. Orbital Models
 - D. Emergy Levels
- 7. Is this an example of ionic bonding?



- A. Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- This is just an orbital model.
- It is a chemical rule of thumb that reflects observation that atoms of main-group elements tend to combine in such a way that each atom has eight electrons in its valence shell, giving it the same electronic configuration as a noble gas.
 - A. Octet Bulle
 - 8. Valence Electrons
 - C. Atomic Orbitals
 - D. Chemical Bonds
- It is the smallest constituent unit of ordinary matter that has the properties of a chemical element.
 - A. Nieutron
 - B. Electron
 - C. Proton
 - D. Atom
- 10. It is a type of chemical bond that involves the electrostatic attraction between oppositely

Grade 11 Physical Science Final Exam Limpopo

SB Merriam

Grade 11	Physical	Science	Final	Exam	Limpopo:

This book delves into Grade 11 Physical Science Final Exam Limpopo. Grade 11 Physical Science Final Exam Limpopo is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Grade 11 Physical Science Final Exam Limpopo, encompassing both the fundamentals and more intricate discussions.

- 1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Grade 11 Physical Science Final Exam Limpopo
 - Chapter 2: Essential Elements of Grade 11 Physical Science Final Exam Limpopo
 - Chapter 3: Grade 11 Physical Science Final Exam Limpopo in Everyday Life
 - Chapter 4: Grade 11 Physical Science Final Exam Limpopo in Specific Contexts
 - \circ Chapter 5: Conclusion
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- 3. In chapter 2, the author will delve into the foundational concepts of Grade 11 Physical Science Final Exam Limpopo. This chapter will elucidate the essential principles that need to be understood to grasp Grade 11 Physical Science Final Exam Limpopo in its entirety.
- 4. In chapter 3, the author will examine the practical applications of Grade 11 Physical Science Final Exam Limpopo in daily life. This chapter will showcase real-world examples of how Grade 11 Physical Science Final Exam Limpopo can be effectively utilized in everyday scenarios.
- 5. In chapter 4, this book will scrutinize the relevance of Grade 11 Physical Science Final Exam Limpopo in specific contexts. The fourth chapter will explore how Grade 11 Physical Science Final Exam Limpopo is applied in specialized fields, such as education, business, and technology.
- 6. In chapter 5, the author will draw a conclusion about Grade 11 Physical Science Final Exam Limpopo. The final chapter will summarize the key points that have been discussed throughout the book.
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