



Department of Provincial Education - NWP

Third Term Test - 2019

Geography - I

Grade 11

Time - 3hours

Name/No:

- This paper consists with 40 questions.
- Answer all the questions using the given guidelines on this paper itself.

● Select the most suitable answer on the given space for question no. 01 to 10.

- 01 The difference between the temperature of day and night is at a higher level can be seen in the (Stratosphere/ Mesosphere/ Thermosphere)
- 02 The extends between the depth from 640Km to 2900Km of the Earth. (upper mantle/ Lower mantle/ Outer core)
- 03 The process of photosynthesis which is necessary for the existence of man occurs in the (Lithosphere/ atmosphere/ Biosphere)
- 04 Latitudinal extent of forests are between 30° - 50° North and South latitude of the Earth. (Tropical/ Temperate/ Coniferous)
- 05 Cultivation of crops like Coffee, Pepper, Betel in Sri Lanka is Known as crops. (estate plantation/ Minor export/ main export)
- 06 Boats produced in Sri Lanka are bought by (South Korea/ Netherland/ United States)
- 07 The highest number of the labour force in Sri Lanka employees in the sector. (Agricultural/ industrial/ service)
- 08 is needed to make high quality porcelain. (Feldspar/ Kaolin/ Clay)
- 09 Central African Sahel zone is mostly affected by in the world. (droughts/ cyclones/ thunderstorms)
- 10 According to 1:50,000 scale, the extent of the mapped area is (80cm x 50cm/ 40cm x 25cm/ 50cm x 25cm)

● If the following statements are correct circle 'T' if not circle 'F' for question no 11 to 15.

- 11 Areas in the southern pole is colder than the northern pole of the earth (T/ F)
- 12 Depressions filled with freshwater located on the land are known as lakes and certain lakes are filled with sea water are considered as inland seas. (T/ F)
- 13 People in the northern part of Sri Lanka use water from tube wells, deep wells, and use ground water for cultivation. (T/ F)
- 14 Cyclones are known as willy - willy in the Pacific Ocean close to the Philippine islands. (T/ F)
- 15 Increase of population at present is higher in the districts in dry zone than the districts in the wet zone in Sri Lanka. (T/ F)

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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a

byproduct of the process as iron ions the iron metal dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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web purpose in this experiment you will observe a single replacement reaction you will also use stoichiometry to predict what the theoretical yield of product is and calculate a percent yield reaction iron copper ii sulfate copper iron ii sulfate pre lab questions write the balanced equation

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web lab report stoichiometry copper sulfate and iron the oxidation of trimethylamine by copper ii chloride nov 10 2022 this research was on the irreversible oxidation of trimethylamine by copper ii chloride which takes place at temperatures of about 75 and higher the coordination of trimethylamine by copper ii

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate CuSO_4 and 2 0153 grams of iron powder Fe the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper's mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web stoichiometric principles can be used to determine the chemical equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction the iron ions dissolve in the copper ii sulfate solution to produce copper ii ions

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away copper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with

a percent yield of 100 72 introduction

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