

QUESTION 7 (Start on a new page.)

# GRADE 11

The equation for the combustion of butane gas is given below.



7.1 Define the term *activation energy*.

part 3

(2)

7.2 Is the combustion reaction of butane *exothermic* or *endothermic*? Give a reason for the answer.

(2)

7.3 Draw a sketch graph of potential energy versus course of reaction for the reaction above.

Clearly indicate the following on the graph:

- Activation energy
- Heat of reaction ( $\Delta H$ )
- Reactants and products

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(3)

7.4 Determine the empirical formula of butane gas if it consists of 82.76% carbon

# Grade11 Chemistry Paper2 2014

**Annelies Wilder-Smith**



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