

Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 3

Section 3.1 Properties of Matter

Label each property as either *physical* or *chemical*.

1. Chemical formula H_2O
2. Forms green carbonate when exposed to moist air
3. Remains unchanged when in the presence of nitrogen
4. Colorless
5. Solid at normal temperatures and pressures
6. Ability to combine with another substance
7. Melting point
8. Liquid at normal temperatures and pressures
9. Boiling point is 100°C
10. Conducts electricity
11. Density is 1.0 g/mL

For each statement below, write *true* or *false*.

12. All matter that we encounter in everyday life exists in one of three physical forms.
13. A solid has definite shape and volume.
14. A liquid has a definite shape and takes on the volume of its container.
15. A gas has both the shape and the volume of its container.
16. The particles in a gas cannot be compressed into a smaller volume.
17. Liquids tend to contract when heated.
18. The particles in a solid are packed far apart.
19. The words *gas* and *vapor* can be used interchangeably.

Section 3.2

What kinds of changes do these words indicate? Write each word under the correct heading. Use each word only once.

	crumble	crush	explode	burn	ferment
Booze					
corrode	rust	condense	melt	oxidize	rot
Physical Change	crumble	crush	condense	melt	oxidize
Chemical Change	corrode	explode	burn	ferment	rot

1. Boil
2. Freeze
3. Condense
4. Evaporate
5. Melt
6. Boil
7. Freeze
8. Condense

For each item in Column A, write the letter of the matching item in Column B.

Column A

Column B

17. The new substances that are formed in a chemical reaction
chemical change a.
18. A chemical reaction that involves one or more substances
reactants b.
- changing into new substances
products c.

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Yan Bai



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