

Chapter Assessment Chemistry: Matter and Change • Chapter 4

Reviewing Vocabulary

Match each definition in Column A with the term in Column B.

Column A

Column B

- | | |
|--|--------------------|
| 1. Radiation deflected toward the positively charged plate | a. atom |
| 2. Atoms with the same number of protons but different numbers of neutrons | b. nucleus |
| 3. High-energy radiation that has no charge and no mass | c. atomic mass |
| 4. The smallest particle of an element that retains the properties of that element | d. |
| 5. The weighted average mass of an element's isotopes | e. gamma ray |
| 6. The center-most part of an atom where the protons and neutrons are contained | f. alpha radiation |
| 7. Radiation deflected toward the negatively charged plate | g. beta |
| 8. The rays and particles emitted by radioactive material | h. atomic |
| 9. Equal to $1/12$ the mass of a carbon-12 atom | i. |
| 10. Ray of radiation traveling from the cathode to the anode | j. Dalton's |
| 11. Process not requiring energy by which unstable nuclei lose energy | k. cathode |
| 12. States that all matter is composed of atoms | l. |
| 13. Process whereby some substances spontaneously emit radiation | m. |
- Compare and contrast each pair of related terms.
14. mass number, atomic number
15. nuclear reaction, nuclear equation

Understanding Main Ideas

Use the periodic table to identify each element described below.

- atomic number 63
- 18 protons
- 44 protons and 44 electrons
- atomic number 24
- 21 protons
- atomic number 55

In the space at the left, write true if the statement is true; if the statement is false,

change the italicized term to make it true.

- An atom's nucleus contains its protons and electrons.
- Neutrons have no electrical charge.
- Beta particles have a charge of 2^+ .
- An alpha particle consists of two protons and two electrons.

Complete the table below.

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Chemistry: Matter & Change, Solving Problems - A Chemistry Handbook McGraw Hill, 2001-08 Glencoe Chemistry Solving Problems A Chemistry Handbook Matter and Change Chemistry McGraw-Hill Staff, 2001-07 Chemistry John S. Phillips, Cheryl Wistrom, 2000 **Chemistry** Laurel Dingrando, McGraw-Hill Staff, Glencoe/McGraw-Hill, 2001-04-01

Glencoe Earth Science Ralph M. Feather, 1999 Earth science is the study of Earth and space It is the study of such things as the transfer of energy in Earth's atmosphere the evolution of landforms patterns of change that cause weather the scale and structure of stars and the interactions that occur among the water atmosphere and land Earth science in this book is divided into four specific areas of study geology meteorology astronomy and oceanography p 8 9 **Resources for**

Teaching Middle School Science Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources

for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents **Glencoe Science** McGraw-Hill Staff,2001-06 **ENC Focus** ,2003 **Mathematics and Science for Students with Special Needs** Eisenhower National Clearinghouse for Mathematics and Science Education,2003 **Standardized Test Practice: Student Edition - Glencoe Chemistry Matter and Change** ,2004 **Supplemental Problems - Glencoe Chemistry Matter and Change** ,2013 **Chemistry: Matter & Change, Study Guide For Content Mastery, Student Edition** McGraw Hill,2001-03-30 Study Guide and Reinforcement Worksheets allow for differentiated instruction through a wide range of question formats There are worksheets and study tools for each section of the text that help teachers track students progress toward understanding concepts Guided Reading Activities help students identify and comprehend the important information in each chapter *Glencoe Chemistry* Glencoe/McGraw-Hill,2002 *Glencoe chemistry* ,2012 *Chemistry Matter and Its Changes Student Solutions Manual* James E. Brady, *Chemistry* Nicholas Drapela,Alison Hyslop,2005 **Chemistry- Matter and Change** , *Chemistry* McGraw-Hill Staff,2001-09-01 *Chemistry* McGraw-Hill Staff,2002-02-01 *Chemistry* McGraw-Hill Staff,2001-09-01

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