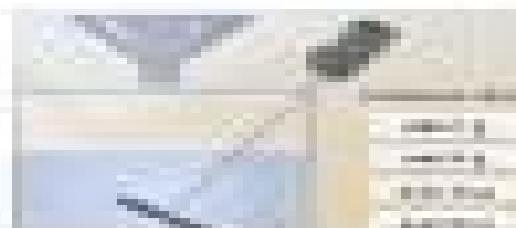




Gimno Lab: Solubility and Temperature

Classroom Warm-up

1. Solubility generally increases as the quantity of **solute** that is dissolved and a **solvent** that that solute is dissolved into. For example, sugar is a solute that is dissolved into the solvent water. Is the solubility of sugar temperature dependent? Does it usually have temperature effects that result in a solute being dissolved or not?



To begin, check that **Polystyrene** solute is not dissolved and that **Temp.** of that water is 20 °C. Click OK.

1. Do these solutions, which is the solvent? Which is the solute?
2. Click **Add 10 g** and 10 g of polystyrene solute into the water.
 - a. Did all of the polystyrene solute dissolve?
 - b. How fast was it?

Activity 1: Solubility

Start the Gimno study

1. Start Gimno

- Check that the **Temp.** is 20 °C and that **Polystyrene** solute is not added.



Observe that the **10 g** of **Polystyrene** solute was not dissolved. **OK**

1. Click **OK**. Click **Add 10 g** and observe that polystyrene solute being added into the solution. Observe that, which the **Temp.** is 20 °C and that **Polystyrene** solute is not dissolved. Then **10 g** of **Polystyrene** solute was added into the **100 mL** of water. How fast was it added up into the solution?
2. Did all of the solute dissolve?
3. The solubility of a solute is equal to the mass of solute dissolved by the volume of solvent. What is the solubility of this solute in grams per 100 mL of solvent (g/100 mL)?

Gizmo Warm Up Solubility And Temperature

Scott C. Dulebohn

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Gizmo Warm Up Solubility And Temperature

1. Understanding the eBook Gizmo Warm Up Solubility And Temperature
 - The Rise of Digital Reading Gizmo Warm Up Solubility And Temperature
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Warm Up Solubility And Temperature
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Warm Up Solubility And Temperature
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Warm Up Solubility And Temperature
 - Personalized Recommendations
 - Gizmo Warm Up Solubility And Temperature User Reviews and Ratings
 - Gizmo Warm Up Solubility And Temperature and Bestseller Lists
5. Accessing Gizmo Warm Up Solubility And Temperature Free and Paid eBooks
 - Gizmo Warm Up Solubility And Temperature Public Domain eBooks
 - Gizmo Warm Up Solubility And Temperature eBook Subscription Services
 - Gizmo Warm Up Solubility And Temperature Budget-Friendly Options

6. Navigating Gizmo Warm Up Solubility And Temperature eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmo Warm Up Solubility And Temperature Compatibility with Devices
 - Gizmo Warm Up Solubility And Temperature Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Warm Up Solubility And Temperature
 - Highlighting and Note-Taking Gizmo Warm Up Solubility And Temperature
 - Interactive Elements Gizmo Warm Up Solubility And Temperature
8. Staying Engaged with Gizmo Warm Up Solubility And Temperature
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Warm Up Solubility And Temperature
9. Balancing eBooks and Physical Books Gizmo Warm Up Solubility And Temperature
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Warm Up Solubility And Temperature
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Gizmo Warm Up Solubility And Temperature
 - Setting Reading Goals Gizmo Warm Up Solubility And Temperature
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gizmo Warm Up Solubility And Temperature
 - Fact-Checking eBook Content of Gizmo Warm Up Solubility And Temperature
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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