Student Exploration: Phase Changes

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: altitude, boil, boiling point, freeze, freezing point, gas, liquid, melt, melting point, phase, solid

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

Marla wants a hard-boiled egg for breakfast. The recipe says boil water, then lower the heat so the water is simmering. But Marla is in a rush, so she keeps the heat on high as the egg cooks.

Do you think the egg will be done faster if the heat is kept on high as the water boils?

Nio

Explain your answer.

If the water is kept high the egg will cook inconsistently so the outer and inner layers will be at different stages.

Gizmo Warm-up

In the Phase Changes Gizmo, select **Micro view** and set the **Ice volume** to 50 cm². Notice the nitrogen (), govgen (), and water () molecules.

To start, create ice. Set Add/remove heat energy to -200 J/s, click Play (), and then click Pause () when the container is about half ice, half water. Set Add/remove heat energy back to 0 J/s, and then click Play.



- Observe water molecules in the solid (ice), liquid (water), and gas (air) phases.
 - A. In which phase(s) do molecules move freely?
 - B. In which phase(s) are molecules held together?
 - C. What do you notice about the positions of the ice molecules?

Gas

Solid

They are all about equal distance apart and more uniform than the air or water molecules.

- Click Reset (2), and select Macro view. Set the ice volume to 25 cm³.
 - A. In which phase(s) is water held in a defined shape?
 - B. In which phase(s) does the water take the shape of the container?

Solid

Liquid

Gizmo Student Exploration Phase Changes Answer Key

Daniela Niemeyer

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