

Ionic Compounds: Names and Formulas Worksheet

1. Write the formulas for the following compounds.

- (a) sodium fluoride NaF
 (b) potassium chloride KCl
 (c) calcium oxide CaO
 (d) lithium bromide LiBr
 (e) beryllium chloride BeCl_2
 (f) potassium sulfide K_2S
 (g) magnesium nitride Mg_3N_2

- (h) calcium phosphide Ca_3P_2
 (i) magnesium oxide MgO
 (j) aluminum chloride AlCl_3
 (k) sodium iodide NaI
 (l) potassium nitride K_3N
 (m) lithium sulfide Li_2S
 (n) barium phosphide Ba_3P_2

2. Write the names for the following compounds.

- (a) LiCl lithium chloride
 (b) MgS Magnesium sulfide
 (c) NaCl sodium chloride
 (d) Al_2O_3 aluminum oxide
 (e) CaS calcium sulfide
 (f) Zn_3P_2 zinc phosphide
 (g) Be_3P_2 beryllium phosphide

- (h) Na_2O sodium oxide
 (i) KBr potassium bromide
 (j) BeO beryllium oxide
 (k) CaF_2 calcium fluoride
 (l) BaBr_2 barium bromide
 (m) Na_3N sodium nitride
 (n) Al_2S_3 aluminum sulfide

Naming Worksheet 5 Ionic Compounds Summary Answer Key

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Introduction to Chemistry, Laboratory Manual T. R. Dickson, 1994-12-23 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students *Hazardous Materials Emergency Response* ,1987

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