

Assignment #2 – Compound Names and Formulas
Multi-valent ions only

A. Write the correct formula for the following compounds, all of which have been named using the modern Roman Numeral Method. The ionic charge is given after the first element (metallic) in Roman Numerals.

1. copper (II) oxide CuO
2. mercury (I) oxide Hg₂O
3. gold (III) chloride AuCl₃
4. thallium (III) bromide TlBr₃
5. bismuth (V) oxide Bi₂O₅
6. terbium (III) oxide Tb₂O₃
7. uranium (VI) oxide UO₃
8. protactinium (IV) bromide PaBr₄
9. cerium (III) oxide Ce₂O₃
10. arsenic (V) sulphide As₂S₅

11. manganese (III) oxide Mn₂O₃
12. vanadium (II) bromide VBr₂
13. niobium(V) oxide Nb₂O₅
14. titanium (III) oxide Ti₂O₃
15. titanium (III) nitride TiN
16. iron (II) oxide FeO
17. cobalt (II) phosphide Co₃P₂
18. tin (II) oxide SnO
19. thulium (II) bromide TmBr₂
20. copper (II) bromide CuBr₂

B. Name the compound using the Roman Numeral Method and balance the ionic charges.

1. SnCl₄ Tin (IV) Chloride
2. BiBr₅ Bismuth (V) Bromide
3. PoO₂ Polonium (IV) Oxide
4. PbI₂ Lead (II) Iodide
5. HgO Mercury (II) Oxide
6. HgCl Mercury (I) Chloride
7. Au₂O₃ Gold (III) Oxide
8. FeCl₂ Iron (II) Chloride

9. PdF₄ Palladium (IV) Fluoride
10. Os₂O₃ Osmium (III) Oxide
11. MoBr₂ Molybdenum (II) Bromide
12. VCl₅ Vanadium (V) Chloride
13. Mn₂O₃ Manganese (III) Oxide
14. CoO Cobalt (II) Oxide
15. Np₂O₃ Neptunium (III) Oxide
16. V₂O₅ Vanadium (V) Oxide

Names And Formulas Worksheet 1

Ying Liu



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Table of Contents Names And Formulas Worksheet 1

1. Understanding the eBook Names And Formulas Worksheet 1
 - The Rise of Digital Reading Names And Formulas Worksheet 1
 - Advantages of eBooks Over Traditional Books
2. Identifying Names And Formulas Worksheet 1
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Names And Formulas Worksheet 1
 - User-Friendly Interface
4. Exploring eBook Recommendations from Names And Formulas Worksheet 1
 - Personalized Recommendations
 - Names And Formulas Worksheet 1 User Reviews and Ratings
 - Names And Formulas Worksheet 1 and Bestseller Lists
5. Accessing Names And Formulas Worksheet 1 Free and Paid eBooks
 - Names And Formulas Worksheet 1 Public Domain eBooks

- Names And Formulas Worksheet 1 eBook Subscription Services
- Names And Formulas Worksheet 1 Budget-Friendly Options
- 6. Navigating Names And Formulas Worksheet 1 eBook Formats
 - ePub, PDF, MOBI, and More
 - Names And Formulas Worksheet 1 Compatibility with Devices
 - Names And Formulas Worksheet 1 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Names And Formulas Worksheet 1
 - Highlighting and Note-Taking Names And Formulas Worksheet 1
 - Interactive Elements Names And Formulas Worksheet 1
- 8. Staying Engaged with Names And Formulas Worksheet 1
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Names And Formulas Worksheet 1
- 9. Balancing eBooks and Physical Books Names And Formulas Worksheet 1
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Names And Formulas Worksheet 1
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Names And Formulas Worksheet 1
 - Setting Reading Goals Names And Formulas Worksheet 1
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Names And Formulas Worksheet 1
 - Fact-Checking eBook Content of Names And Formulas Worksheet 1
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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