

Answer all the questions.

15. Sir Humphrey Davy first reported the existence of the element sodium (Na), and the first person to isolate the element potassium (K) was Humphrey, the brother of the famous Davy. Both were responsible for the discovery of the element rubidium. Rubidium and potassium cannot melt to each other in the Periodic Table.

A sample of rubidium was weighed and found to consist of two isotopes: rubidium-85 and rubidium-87. Information about these isotopes is given in the table.

Isotope	relative isotopic mass	abundance (%)
rubidium-85	85.468	72.174
rubidium-87	87.471	27.826

- (a) Calculate the relative atomic mass, A_r , of the element rubidium. Give your answer to two decimal places.

_____ [2]

- (b) Define the term relative atomic mass.

_____ [2]

- (c) Calculate the relative atomic mass of the sample of rubidium.

Give your answer to two decimal places.

_____ [2]

Answer to _____ [1]

- (d) Both rubidium, a Group 1 element, and strontium, a Group 2 element, have their outermost electrons in the 5s orbital.

What 3-2-1 shape is then to be expected?

_____ [1]

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