

HONORS PHYSICAL SCIENCE

FINAL EXAM STUDY GUIDE

- Science Skills
 - Explain how science and technology are related
 - Describe the steps of the scientific method
 - Compare and contrast accuracy and precision
 - Perform calculations involving scientific notation and both temperature and metric conversion factors
 - Identify the relationship between a manipulated (independent) variable and a responding (dependent) variable
 - Compare and contrast a scientific law and a scientific theory
- Properties and States of Matter
 - Compare and contrast the characteristics of elements, compounds, pure substances, mixtures (heterogeneous and homogeneous), solutions, suspensions, and colloids
 - Understand the methods of separation
 - Compare and contrast physical and chemical properties, and physical and chemical changes
 - Explain the behavior of gases, liquids, and solids using kinetic theory
 - Predict changes in gas pressure due to changes in temperature, volume, and number of particles
 - Explain Charles's Law, Boyle's Law, and the combined gas law
 - Apply gas laws to solve problems involving gases
 - Name and describe each of the six phase changes. Which are endothermic and which are exothermic?
- Atoms, Elements, and Bonds
 - Compare and contrast Bohr and Rutherford's atomic models
 - Identify three subatomic particles and compare their properties
 - Compare and contrast atoms in ground and excited states
 - Distinguish the atomic number of an element from the mass number of an isotope, and use these numbers to describe the structure of atoms
 - Explain how the electron cloud model represents the behavior and locations of electrons in atoms
 - Understand the development and organization of the periodic table, especially in terms of group properties, atomic mass and valence electrons
 - Identify general properties of metals, nonmetals, and metalloids
 - Predict the reactivity of some elements based on their locations in the periodic table
 - Compare and contrast ionic and covalent bonds
 - Understand the concept of polarity and the role it plays in bonding
- Chemical Reactions, Acids, and Bases
 - Interpret chemical equations in terms of reactants, products, and conservation of mass
 - Balance chemical equations by manipulating coefficients
 - Convert between moles and mass of a substance using molar mass
 - Calculate amounts of reactants or products by using molar mass, mole ratios, and balanced chemical equations
 - Classify chemical reactions as synthesis, decomposition, single replacement, double replacement, or combustion reactions
 - Explain how energy is conserved during chemical reactions
 - Describe the factors affecting chemical reaction rates
 - Describe the factors affecting chemical equilibrium (see Chapter's Principles)
 - Define acid and describe some of the general properties of an acid
 - Define base and define some of the general properties of a base
 - Understand what a buffer is and how it can be used
 - Know what a neutralization reaction is, and what the products are
- Nuclear Chemistry
 - Describe the process of nuclear decay

Honors Physical Science Final Review

Terry C. Jones



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Reviewing **Honors Physical Science Final Review**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Honors Physical Science Final Review**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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