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2010 Honors Chemistry Semester 1 Practice Test

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| | | | |

Identify the choice that best completes the statement or answers the question.

- 1. The two most important properties of all matter are
 - a. the ability to carry an electric current well and to hold electric charge.
 - b. taking up space and having mass.
 - c. being brittle and hard.
 - d. being malleable and ductile.
- A compound is
 - a. a pure substance that cannot be broken down into simpler, stable substances.
 - a substance, made of two or more atoms that are chemically bonded, that can be broken down into simpler, stable substances.
 - c. the smallest unit of matter that maintains its chemical identity.
 - d. any substance, whether it is chemically bonded or not.
- Matter includes all of the following except
 - m. mir.

e. samodoe.

b. light.

- d. water vapor.
- A student recorded the following while completing an experiment.

Color of substance: yellow, shiny powder

Effect of magnet: yellow, shiny powder was attracted

The student should classify the substance as a(n)

a. element.

e. missture.

b. compound.

- d. plasma.
- 5. Which of the following is not a physical change?
 - a. grinding

boiling

b. cutting

d. burning

- A physical change occurs when a
 - a. peach spoils.
 - b. silver bowl tarnishes.
 - c. bracelet turns your wrist green.
 - d. glue gun melts a glue stick.

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