Chemical Reaction Example Oxided Practice Problems

Directions: Write balanced chemical equations for each of the following reactions. Be sure to include phases of matter.

- 1. Solid iron (III) oxide reacts with gaseous hydrogen chloride to form iron (III) chloride and water vapor.
- Nitric acid dissolved in water reacts with solid sodium carbonate to form liquid water and carbon dissolved gas and sodium nitrate dissolved in water.

Convert these equations to sentences

- 3. $Fe_{(r)}$ + $O_{2(g)}$ \rightarrow $Fe_2O_{3(r)}$
- 4. $Cu_{(0)}$ + $AgNO_{X(N)}$ \rightarrow $Ag_{(0)}$ + $Cu(NO_1)_{X(N)}$
- NO_{3(d)} → N_{3(d)} + O_{3(d)}

Additional Practice

- Silver nitrate solution and potassium chromate solution, when mixed together, produce solid silver chromate and potassium nitrate solution.
- Phosphoric acid solution reacts with barium chloride solution to produce solid barium phosphate and hydrochloric acid solution.
- 8. Solid magnesium burns in air to produce magnesium oxide, a solid white powder.
- 9. When solid sulfur is exposed to gaseous fluorine, sulfur hexafluoride gas is formed.
- When exposed to manganese (IV) oxide, hydrogen peroxide rapidly breaks down into water and oxygen gas.
- Solid sodium carbonate reacts with hydrochloric acid solution to produce carbon dioxide gas, water and sodium chloride solution.

Guided Practice Problems Chemistry Ch 16

Disha Experts

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This book delves into Guided Practice Problems Chemistry Ch 16. Guided Practice Problems Chemistry Ch 16 is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Guided Practice Problems Chemistry Ch 16, encompassing both the fundamentals and more intricate discussions.

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