Name _____ Chemistry I-2 HONORS Study Guide for Acids and Bases (Chapter 17)

Students should be able to

- Know properties of acids and bases
- Know definitions of acids and bases according to Arrhenius, Bronsted-Lowry, and Lewis.
- Label Bronsted-Lowry acid-base reactions as acid, base, conjugate acid, and conjugate base
- Identify polyprotic acids and bases.
- Review naming of acids, both binary and ternary.
- Calculate pH, pOH, H+ concentration, and OH- concentration for either a strong or weak acid or base.
- Calculate the solute and solvent necessary for a dilution
- Calculate the <u>volume</u> necessary to neutralize an acid/base
- Measure the pH and acidity/alkalinity of household chemicals (laboratory)
- Calculate the % by mass of acetic acid in vinegar (laboratory)

Review problems

- 1. List three properties of acids.
- List three properties of bases.

Fill in the table concerning definitions of acids and bases.

Definition	Acid	Base
Traditional (Arrhenius)		
	Electron pair acceptor	
		Proton acceptor

Name the following acids. Circle any polyprotic acids.

- 9. H₂PO₄_____
- 10. HF
- 11. H₂S

Solve.

14. [H+] = 5.32 X 10⁻⁴ [OH-] = pH = pOH = Acid or Base?

Guided And Study Workbook Acids And Bases

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Guided And Study Workbook Acids And Bases:

Guided And Study Workbook Acids And Bases Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has be apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Guided And Study Workbook Acids And Bases**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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