

## Unit

# 1

## SOME BASIC CONCEPTS OF CHEMISTRY

### I. Multiple Choice Questions (Type-I)

1. Two students performed the same experiment separately and each one of them recorded two readings of mass which are given below. Correct reading of mass is 3.0 g. On the basis of given data, mark the correct option out of the following statements.
- | Student | Readings |      |
|---------|----------|------|
|         | (i)      | (ii) |
| A       | 3.01     | 2.99 |
| B       | 3.05     | 2.95 |
- (i) Results of both the students are neither accurate nor precise.  
(ii) Results of student A are both precise and accurate.  
(iii) Results of student B are neither precise nor accurate.  
(iv) Results of student B are both precise and accurate.
2. A measured temperature on Fahrenheit scale is  $200^{\circ}\text{F}$ . What will this reading be on Celsius scale?
- (i)  $40^{\circ}\text{C}$   
(ii)  $94^{\circ}\text{C}$   
(iii)  $93.3^{\circ}\text{C}$   
(iv)  $30^{\circ}\text{C}$
3. What will be the molarity of a solution, which contains 3.85 g of  $\text{NaCl(s)}$  per 500 mL?
- (i)  $4\text{ mol L}^{-1}$   
(ii)  $20\text{ mol L}^{-1}$

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**Zhe Xu**



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