

CHAPTER 21 REVIEW

Nuclear Chemistry**SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

- _____ Based on the information about the three elementary particles in the text, which has the greatest mass?
 - the proton
 - the neutron
 - the electron
 - They all have the same mass.
- _____ The force that keeps nucleons together is
 - a strong nuclear force.
 - a weak nuclear force.
 - an electromagnetic force.
 - a gravitational force.
- _____ The stability of a nucleus is most affected by the
 - number of neutrons.
 - number of protons.
 - number of electrons.
 - ratio of neutrons to protons.
- _____ If an atom should form from its constituent particles,
 - matter is lost and energy is taken in.
 - matter is lost and energy is released.
 - matter is gained and energy is taken in.
 - matter is gained and energy is released.
- _____ For atoms of a given mass number, those with greater mass defects, have
 - smaller binding energies per nucleon.
 - greater binding energies per nucleon.
 - the same binding energies per nucleon as those with smaller mass defects.
 - variable binding energies per nucleon.
- Based on **Figure 1.1** of the text, which isotope of He, helium-3 or helium-4,
 - _____ a. has the smaller binding energy per nucleon?
 - _____ b. is more stable to nuclear changes?
- The number of neutrons in an atom of magnesium-25 is _____.
- Nuclides of the same element have the same _____.

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Table of Contents Glencoe Chemistry Chapter 13 Answer

1. Understanding the eBook Glencoe Chemistry Chapter 13 Answer
 - The Rise of Digital Reading Glencoe Chemistry Chapter 13 Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Glencoe Chemistry Chapter 13 Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Glencoe Chemistry Chapter 13 Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Glencoe Chemistry Chapter 13 Answer
 - Personalized Recommendations
 - Glencoe Chemistry Chapter 13 Answer User Reviews and Ratings
 - Glencoe Chemistry Chapter 13 Answer and Bestseller Lists
5. Accessing Glencoe Chemistry Chapter 13 Answer Free and Paid eBooks
 - Glencoe Chemistry Chapter 13 Answer Public Domain eBooks
 - Glencoe Chemistry Chapter 13 Answer eBook Subscription Services
 - Glencoe Chemistry Chapter 13 Answer Budget-Friendly Options
6. Navigating Glencoe Chemistry Chapter 13 Answer eBook Formats

- ePub, PDF, MOBI, and More
 - Glencoe Chemistry Chapter 13 Answer Compatibility with Devices
 - Glencoe Chemistry Chapter 13 Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Glencoe Chemistry Chapter 13 Answer
 - Highlighting and Note-Taking Glencoe Chemistry Chapter 13 Answer
 - Interactive Elements Glencoe Chemistry Chapter 13 Answer
 8. Staying Engaged with Glencoe Chemistry Chapter 13 Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Glencoe Chemistry Chapter 13 Answer
 9. Balancing eBooks and Physical Books Glencoe Chemistry Chapter 13 Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Glencoe Chemistry Chapter 13 Answer
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Glencoe Chemistry Chapter 13 Answer
 - Setting Reading Goals Glencoe Chemistry Chapter 13 Answer
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Glencoe Chemistry Chapter 13 Answer
 - Fact-Checking eBook Content of Glencoe Chemistry Chapter 13 Answer
 - Distinguishing Credible Sources
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 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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