

Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 3

Section 3.1 Properties of Matter

Label each property as either *physical* or *chemical*.

1. Chemical formula H_2O
2. Forms green carbonate when exposed to moist air
3. Remains unchanged when in the presence of nitrogen
4. Colorless
5. Solid at normal temperatures and pressures
6. Ability to combine with another substance
7. Melting point
8. Liquid at normal temperatures and pressures
9. Boiling point is 100°C
10. Conducts electricity
11. Density is 1.0 g/mL

For each statement below, write *true* or *false*.

12. All matter that we encounter in everyday life exists in one of three physical forms.
13. A solid has definite shape and volume.
14. A liquid has a definite shape and takes on the volume of its container.
15. A gas has both the shape and the volume of its container.
16. The particles in a gas cannot be compressed into a smaller volume.
17. Liquids tend to contract when heated.
18. The particles in a solid are packed far apart.
19. The words *gas* and *vapor* can be used interchangeably.

Section 3.2

What kinds of changes do these words indicate? Write each word under the correct heading. Use each word only once.

	crumple	crush	explode	burn	ferment
Booze					
corrode	rust	laminar	vaporize	oxidize	rot
Physical Change					
Chemical Change					

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____
8. _____

For each item in Column A, write the letter of the matching item in Column B.

Column A

Column B

- | | |
|---------------------------------------------------------------|----|
| 17. The new substances that are formed in a chemical reaction | a. |
| chemical change | |
| 18. A chemical reaction that involves one or more substances | b. |
| reactants | |
| changing into new substances | c. |
| products | |

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Marco Cascella



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Table of Contents Glencoe Chemistry Study Guide Answers

1. Understanding the eBook Glencoe Chemistry Study Guide Answers
 - The Rise of Digital Reading Glencoe Chemistry Study Guide Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Glencoe Chemistry Study Guide Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Glencoe Chemistry Study Guide Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Glencoe Chemistry Study Guide Answers
 - Personalized Recommendations
 - Glencoe Chemistry Study Guide Answers User Reviews and Ratings
 - Glencoe Chemistry Study Guide Answers and Bestseller Lists
5. Accessing Glencoe Chemistry Study Guide Answers Free and Paid eBooks
 - Glencoe Chemistry Study Guide Answers Public Domain eBooks
 - Glencoe Chemistry Study Guide Answers eBook Subscription Services
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6. Navigating Glencoe Chemistry Study Guide Answers eBook Formats
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 - Glencoe Chemistry Study Guide Answers Compatibility with Devices
 - Glencoe Chemistry Study Guide Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Glencoe Chemistry Study Guide Answers
 - Highlighting and Note-Taking Glencoe Chemistry Study Guide Answers
 - Interactive Elements Glencoe Chemistry Study Guide Answers
8. Staying Engaged with Glencoe Chemistry Study Guide Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Glencoe Chemistry Study Guide Answers
9. Balancing eBooks and Physical Books Glencoe Chemistry Study Guide Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Glencoe Chemistry Study Guide Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Glencoe Chemistry Study Guide Answers
 - Setting Reading Goals Glencoe Chemistry Study Guide Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Glencoe Chemistry Study Guide Answers
 - Fact-Checking eBook Content of Glencoe Chemistry Study Guide Answers
 - Distinguishing Credible Sources
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 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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