

## Stoichiometry Study Guide

Students should be able to...

- Explain mole ratio
- Convert moles, grams, atoms/molecules
- Calculate theoretical yields when given the limiting reactant (Convert: moles A to moles B, grams A to moles B, moles A to grams B, grams A to grams B)
- Explain limiting reactant and excess.
- Calculate theoretical yields by finding the limiting reactant
- Calculate percent yield.
- Calculate percent composition.
- Calculate theoretical yield, limiting reactant, excess, and percent yield for a double replacement reaction (laboratory)

### Practice Problems

Find the molar mass of each compound.

1.  $\text{BaI}_2$
2.  $\text{AgC}_2\text{H}_3\text{O}_2$

Convert.

3. 0.4 moles of Sn to atoms
4. 32 g of  $\text{K}_2\text{O}$  to molecules
5. 3.7 g of  $\text{H}_2$  to moles
6. 800 g of  $\text{CaCl}_2$  to moles
7. 0.75 moles of  $\text{H}_2\text{O}$  to grams
8. 9.0 g of  $\text{Cl}_2$  to atoms
9. 0.5 moles of  $\text{KNO}_3$  to grams

Find the percent composition for each element in the compound.

10.  $\text{Mn}_2\text{O}_7$
11.  $\text{AuNO}_3$

### Stoichiometry



If you started with 3.7 moles of magnesium oxide, how many moles of magnesium hydroxide would be produced?



If 2.8 grams of sodium were used for the reaction, and an excess of iodine was present, how many moles of sodium iodide would be produced?



# Glencoe Chemistry Stoichiometry Study Guidekey

**Charles H. Corwin, Pearson**



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Edward C. Stark,1979      Introductory Chemistry Study Guide Holder,2003-05      **Study Guide** Darrell D. Ebbing,Susan M. Schelble,1998      **The Key Student Study Guide** ,2023      **Organic Chemistry Study Guide with Solutions Manual** Neil E. Schore,2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer      Chemistry and Student Study Guide and Study Guide, Fifth Edition and Smarthinking and Letter Zumdahl,2002-01-01      Glencoe Chemistry: Concepts and Applications, Study Guide McGraw-Hill,1996-06      **The Key Student Study Guide** ,2008      **Study Guide [to Accompany] General Chemistry** James E. Brady,1982      *Study Guide, Introductory Chemistry, a Foundation ; Introductory Chemistry ; Basic Chemistry* Iris Stovall,Donald J. DeCoste,2000      *The Key* Rao Gautam,2003      **Supplement** Charles H. Corwin,Pearson,2004-07

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