Chapter Assessment Chemistry: Matter and Change • Chapter 4 Reviewing Vocabulary Match each definition in Column A with the term in Column II. Collegemen & Continuous St. Radiation deflected toward the positively changed plate. and the second Atoms with the same number of protons but different numbers the involved about of neputrons es, estornic myeres. 3. High-energy radiation that has no charge and no mass - 企業の表現の企業を 4. The smallest particle of an element that retains the properties ALCOHOLOGY CO. Company of of that ellerated d. Michael radiation. 5. The weighted average mass of an element's isotopes en bette PARTICIPATION OF THE PROPERTY. 6. The center most part of an atom where the protons and neutrons they let connict PERSONAL PROPERTY. serve is contrained. ш PROBLEMS OF Radiation deflected toward the negatively charged plate. Distribution is MEDICAL C. STREET, CANAL. The rays and particles emitted by redioactive material. the or antiborophics Found to 1712 the mean of a carbon 32 atom. medical action (by) 10. Ray of radiation traveling from the cathode to the anode medical estable discourt 11. Process (not requiring energy) by which unstable nuclei loss energy 1.2. Photos that all matter is convenied of atoms. 13. Process whereby some substances spontaneously emit radiation Compare and contrast each pair of related terms. THE PROPER PROPERTY ABOVES CONTRACT 15. nuclear reaction, nuclear equation: Understanding Main Ideas Use the periodic table to identify each element described below. abornic marriboni 65. 26. To protection 3. 44 protons and 44 electrons.

- 4. adjornic reanniser 24
- No. 273. Introduction
- abornic number 55.
- In the space at the left, write draw if the statement is true: if the statement is false.

change the italicized term to make it true.

- Y. An aborn's nucleus contains its protons and electrons.
- the Manufacture have not effect the all change.
- dieta participa have a charge of 2°
- 10. An alpha particle consists of two protons and two electrons.

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