

# COVALENT BONDING

Name \_\_\_\_\_

Covalent bonding occurs when two or more nonmetals share electrons, attempting to attain a stable octet of electrons at least part of the time. For example:

*H wants 2 electrons  
all other atoms 8*



Note that hydrogen is content with 2, not 8, electrons.

*electrons in the middle share*

Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons.

*wanted available share*

<p>1. <math>\text{H} + \text{H} (\text{H}_2)</math></p> <p>W - 4</p> <p>A - 2</p> <p>S - 2</p>	<p><i>wanted 2 + 2</i></p> <p><i>available</i></p> <p><math>\text{H} \cdot \cdot \text{H}</math></p> <p><math>\text{H} \cdot \quad \cdot \text{H}</math></p>	<p><math>\text{H} - \text{H}</math></p> <p><i>single bond = share 2 e<sup>-</sup></i></p>
<p>2. <math>\text{F} + \text{F} (\text{F}_2)</math></p> <p>W - 8 + 8 = 16</p> <p>A - 7 + 7 = 14</p> <p>S - 2</p>	<p><math>\cdot \ddot{\text{F}} \cdot \cdot \ddot{\text{F}} \cdot</math></p> <p><i>2 e<sup>-</sup></i></p>	<p><math>\cdot \ddot{\text{F}} - \ddot{\text{F}} \cdot</math></p> <p><i>single bond 2 e<sup>-</sup></i></p>
<p>3. <math>\text{O} + \text{O} (\text{O}_2)</math></p> <p>W - 8 + 8 = 16</p> <p>A - 6 + 6 = 12</p> <p>S - 4</p>	<p><math>\cdot \ddot{\text{O}} \cdot \cdot \ddot{\text{O}} \cdot</math></p> <p><i>4 e<sup>-</sup></i></p>	<p><math>\cdot \ddot{\text{O}} = \ddot{\text{O}} \cdot</math></p> <p><i>4 double bond</i></p>
<p>4. <math>\text{N} + \text{N} (\text{N}_2)</math></p> <p>W - 8 + 8 = 16</p> <p>A - 5 + 5 = 10</p> <p>S - 6</p>	<p><math>\cdot \ddot{\text{N}} \cdot \cdot \ddot{\text{N}} \cdot</math></p>	<p><math>\cdot \ddot{\text{N}} \equiv \ddot{\text{N}} \cdot</math></p>
<p>5. <math>\text{C} + \text{O} (\text{CO}_2)</math></p> <p>W - 8 + 8 + 8 = 24</p> <p>A - 4 + 6 + 6 = 16</p> <p>S - 8</p>	<p><math>\cdot \ddot{\text{O}} \cdot \cdot \ddot{\text{C}} \cdot \cdot \ddot{\text{O}} \cdot</math></p>	<p><math>\cdot \ddot{\text{O}} = \text{C} = \ddot{\text{O}} \cdot</math></p>
<p>6. <math>\text{H} + \text{O} (\text{H}_2\text{O})</math></p> <p>W - 2 + 2 + 8 = 12</p> <p>A - 1 + 1 + 6 = 8</p> <p>S - 4</p>	<p><math>\text{H} \cdot \cdot \ddot{\text{O}} \cdot \cdot \text{H}</math></p>	<p><math>\text{H} - \ddot{\text{O}} - \text{H}</math></p>

# Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key

**Minjie Lin**



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## **Table of Contents Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key**

1. Understanding the eBook Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
  - The Rise of Digital Reading Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
  - User-Friendly Interface

4. Exploring eBook Recommendations from Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
  - Personalized Recommendations
  - Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key User Reviews and Ratings
  - Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key and Bestseller Lists
5. Accessing Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key Free and Paid eBooks
  - Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key Public Domain eBooks
  - Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key eBook Subscription Services
  - Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key Budget-Friendly Options
6. Navigating Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key Compatibility with Devices
  - Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
  - Highlighting and Note-Taking Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
  - Interactive Elements Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
8. Staying Engaged with Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
9. Balancing eBooks and Physical Books Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
  - Setting Reading Goals Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
  - Carving Out Dedicated Reading Time



12. Sourcing Reliable Information of Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
  - Fact-Checking eBook Content of Glencoe Chemistry Chapter 9 Covalent Bonding Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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