

Student Exploration: Temperature and Particle Motion

Essential Question: How do the temperatures of gases reflect the kinetic energy of the particles? How do the temperatures of liquids reflect the kinetic energy of the particles?

<p>Learning GOALS:</p> <p>Knowledge:</p>	<p>Learning GOALS:</p> <p>1. Explain how the temperature of a gas is related to the kinetic energy of the particles.</p>	
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Question: How do the temperatures of gases reflect the kinetic energy of the particles?

- Collect:** Use the Temperature and Particle Motion simulation. Consider the particles motion at 100 K.
- Analyze:** The temperature of a substance is a measure of the average kinetic energy of the particles (translational, vibrational, and rotational). The kinetic energy of a particle is proportional to its mass and the square of its velocity.
 - Based on the formula for kinetic energy, how will the temperature change if you increase the average velocity of the molecules ten fold?
 - How will the temperature change if you increase the mass of the gas molecules?
- Apply:** Oxygen molecules are diatomic molecules (two oxygen atoms). At the same temperature, nitrogen gas molecules would have a higher average velocity than oxygen molecules.
- Collect:** Use your simulation to measure the average velocity of the particles at 100 K.
- Apply:** Based on the definition of temperature, explain why oxygen molecules move faster than nitrogen molecules at the same temperature.

<p>Learning GOALS:</p> <p>Knowledge:</p>	<p>Learning GOALS:</p> <p>1. Explain how the temperature of a gas is related to the kinetic energy of the particles.</p>	
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Question: How do the temperatures of liquids reflect the kinetic energy of the particles? How do the temperatures of liquids reflect the kinetic energy of the particles?

Question: How do the temperatures of liquids reflect the kinetic energy of the particles?

- Collect:** Use the Temperature and Particle Motion simulation. Use the data to create the graph of the average velocity of the particles at 100 K.

Gizmo Temperature And Particle Motion Answers

RJ Shavelson

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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