

Name: _____

Date: _____

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? **They can share the markers.**

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

They can give the ones they need to each other.

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** (▶) to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? _____7_____
2. Click **Pause** (⏸). Drag an electron from the left atom to the right atom. Click **Play**.
What happens? **They share an electron.**
3. Click **Pause**, drag an electron from the right atom to the left, and then click **Play**.
What happens now? **They share another one.**

Gizmo Covalent Bonds Answer Key

Julia Schneider

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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Table of Contents Gizmo Covalent Bonds Answer Key

1. Understanding the eBook Gizmo Covalent Bonds Answer Key
 - The Rise of Digital Reading Gizmo Covalent Bonds Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Covalent Bonds Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Covalent Bonds Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Covalent Bonds Answer Key
 - Personalized Recommendations
 - Gizmo Covalent Bonds Answer Key User Reviews and Ratings
 - Gizmo Covalent Bonds Answer Key and Bestseller Lists
5. Accessing Gizmo Covalent Bonds Answer Key Free and Paid eBooks
 - Gizmo Covalent Bonds Answer Key Public Domain eBooks
 - Gizmo Covalent Bonds Answer Key eBook Subscription Services
 - Gizmo Covalent Bonds Answer Key Budget-Friendly Options

6. Navigating Gizmo Covalent Bonds Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmo Covalent Bonds Answer Key Compatibility with Devices
 - Gizmo Covalent Bonds Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Covalent Bonds Answer Key
 - Highlighting and Note-Taking Gizmo Covalent Bonds Answer Key
 - Interactive Elements Gizmo Covalent Bonds Answer Key
8. Staying Engaged with Gizmo Covalent Bonds Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Covalent Bonds Answer Key
9. Balancing eBooks and Physical Books Gizmo Covalent Bonds Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Covalent Bonds Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Gizmo Covalent Bonds Answer Key
 - Setting Reading Goals Gizmo Covalent Bonds Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gizmo Covalent Bonds Answer Key
 - Fact-Checking eBook Content of Gizmo Covalent Bonds Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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