

Student Exploration: Ionic Bonds

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

Nate can give Clara a marker

2. Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

Maggie can give Fred and Florence one marker each

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The Ionic Bonds Gizmo™ allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** (▶) to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



1. Each atom consists of a central nucleus and several **shells** that contain electrons. The outermost electrons are called **valence electrons**. (Inner electrons are not shown.)

How many valence electrons does each atom have?

Sodium: 1

Chlorine: 7

2. Click Pause (⏸). Elements can be classified as **metals** and **nonmetals**. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. **Electron affinity** is a measure of how tightly the valence electrons are held.

A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

sodium

Which is a nonmetal?

chlorine

B. Try moving an electron from the metal to the nonmetal. What happens?

The chlorine atom took the electron and added it to its outermost shell

Gizmo Answer Key Ionic Bonds

Sabine Zange

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Gizmo Answer Key Ionic Bonds

1. Understanding the eBook Gizmo Answer Key Ionic Bonds
 - The Rise of Digital Reading Gizmo Answer Key Ionic Bonds
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Answer Key Ionic Bonds
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Answer Key Ionic Bonds
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Answer Key Ionic Bonds
 - Personalized Recommendations
 - Gizmo Answer Key Ionic Bonds User Reviews and Ratings
 - Gizmo Answer Key Ionic Bonds and Bestseller Lists
5. Accessing Gizmo Answer Key Ionic Bonds Free and Paid eBooks
 - Gizmo Answer Key Ionic Bonds Public Domain eBooks
 - Gizmo Answer Key Ionic Bonds eBook Subscription Services
 - Gizmo Answer Key Ionic Bonds Budget-Friendly Options

6. Navigating Gizmo Answer Key Ionic Bonds eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmo Answer Key Ionic Bonds Compatibility with Devices
 - Gizmo Answer Key Ionic Bonds Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Answer Key Ionic Bonds
 - Highlighting and Note-Taking Gizmo Answer Key Ionic Bonds
 - Interactive Elements Gizmo Answer Key Ionic Bonds
8. Staying Engaged with Gizmo Answer Key Ionic Bonds
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Answer Key Ionic Bonds
9. Balancing eBooks and Physical Books Gizmo Answer Key Ionic Bonds
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Answer Key Ionic Bonds
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Gizmo Answer Key Ionic Bonds
 - Setting Reading Goals Gizmo Answer Key Ionic Bonds
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gizmo Answer Key Ionic Bonds
 - Fact-Checking eBook Content of Gizmo Answer Key Ionic Bonds
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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