

# CHM 2210 Lab: Melting Point

## Introduction

The melting point of a solid is the temperature at which the solid melts (changes to a liquid) at 1 atm of pressure. Melting point is determined by slow heating of a small solid sample placed in a capillary tube. The melting point should be reported as a melting point range: from the temperature at which the first droplet of liquid is observed to the temperature at which the sample becomes a clear liquid, for example: 96.7 – 98.9°C.

A pure solid (pure enough for most purposes) usually has a sharp melting point which means that it melts within a range of 2°C or less.

An impure solid melts at a broader range (3°C or more) and at depressed temperature.

Using melting point to find the identity of a solid is not sufficient because there are too many substances with the same melting points.

However, you can use a mixed melting point as a tool for identification of a solid.

If you mix two different solids with the same melting point, the melting point of the mixture will be broad and depressed. The melting point of a solid and a mixture of this solid with the authentic solid should be the same.

Example:

Measured/Observed melting point of an unknown solid X: 128.0 – 129.5°C

Melting point of known solid A: 130°C

Melting point of known solid B: 158°C

Measured melting point of mixture X + A: 127.5 – 129.0°C

Measured melting point of mixture X + B: 133.5 – 142.0°C

Identity of unknown X: solid A

In the future you will measure melting points of solids mostly to determine their purity.

- **Preparation of Sample**

A solid sample should be dried and crushed with a metal spatula or bottom of a test tube.

If a solid is used for mixed melting point, the 1:1 mixture should be crushed together into a homogeneous mixture.

The pulverized solid should be mounded up and a small (1 – 2 mm) plug of sample should be pushed into the open end of the capillary tube. The sample should be knocked down to the sealed end of the capillary tube by tapping its sealed end on the surface or dropping it into a long glass tubing.

The sample should be 1-2 mm long and packed firmly.

If the sample is too large, the measured melting point may be too high.

If the sample is loosely packed, it may be difficult to determine the melting point range because of the air present.

# Melting Point Lab For High School

**Elliott Cassidy**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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*Fostering Sustained Learning Among Undergraduate Students: Emerging Research and Opportunities* Chemosit, Caroline, Rugutt, John, Rugutt, Joseph K., 2017-03-03 Keeping students engaged and receptive to learning can at times be a challenge However by the implementation of new methods and pedagogies instructors can strengthen the drive to learn among their students *Fostering Sustained Learning Among Undergraduate Students Emerging Research and Opportunities* is an essential publication for the latest scholarly information on methods to inculcate student learning with a focus on implications to institutional policy and practices Featuring coverage on topics such as financial aid student motivation and mentorship this book is ideally designed for academicians practitioners and researchers seeking novel perspectives on the learning process and instruction methods **New Coordinated Science: Chemistry Students' Book** RoseMarie Gallagher, R. Gallagher, Paul Ingram, 2001-07-19 Provides information in manageable chunks which is reinforced by questions and activities that encourage students to consider the practical application of science to everyday life This work is useful for Higher Tier GCSE students Gourmet Lab Sarah Reeves Young, 2011 Hands on inquiry based and relevant to every student OCOs life Gourmet Lab serves up a full menu of activities for science teachers of grades 6 OCO 12 This collection of 15 hands on experiments OCO each of which includes a full set of both student and teacher pages OCO challenges students to take on the role of scientist and chef as they boil bake and toast their way to better understanding of science concepts from chemistry biology and physics By cooking edible items such as pancakes and butterscotch students have the opportunity to learn about physical changes in states of matter acids and bases biochemistry and molecular structure The Teacher pages include Standards addressed in each lab a vocabulary list safety protocols materials required procedures data analysis student questions answer key and conclusions and connections to spur wrap up class discussions Cross curricular notes are also included to highlight the lesson OCOs connection to subjects such as math and literacy Finally optional extensions for both middle school and high school levels detail how to explore each concept further What better topic than food to engage students to explore science in the natural world Catalogue of the Ann Arbor High School for the Academic Year Ann Arbor High School (Mich.), 1903 *Chemistry* Dorin, Prentice Hall Dictionary Editors, 1992 Conquering the Electron Derek Cheung, Eric Brach, 2020-02-24 *Conquering the Electron* offers readers a true and engaging history of the world of electronics beginning with the discoveries of static electricity and magnetism and ending with the creation of the smartphone and the iPad This book shows the interconnection of each advance to the next on the long journey to our modern day technologies Exploring the combination of genius infighting and luck that powered the creation of today s electronic age *Conquering the Electron* debunks the hero worship so often plaguing the stories of great advances Want to know how AT T s Bell Labs developed semiconductor technology and how its leading scientists almost came to blows in the process Want to understand how radio and television work and why RCA drove their inventors to financial ruin and early graves *Conquering*

the Electron offers these stories and more presenting each revolutionary technological advance right alongside blow by blow personal battles that all too often took place      **Michigan School Moderator** ,1908      *The NIH Record* ,1990      **The Internet of Things for Education** Linda Daniela,2021-12-02 This book is about the Internet of Things in the field of education Specifically it focuses on two major topics IoT Internet of Things solutions to support distance education and new pedagogical approaches to support development of computational thinking with educational devices possessing the characteristics of IoT As the educational landscape has dramatically changed in times of global pandemic online resources and media such as IoT have become increasingly important This situation compels all educational scholars researchers and practitioners to search for new solutions new educational pathways and new agents for knowledge development to support learning This book presents the possibilities of IoT as both a catalyst and performance tool for education The convergence of multiple technologies real time analytics machine learning commodity sensors and embedded systems can serve as tools for learning support and this book details exactly how these powerful tools can be utilized to best effect      **Nuclear Science Abstracts** ,1972      **The United States Catalog** ,1909      AFOSR. United States. Air Force. Office of Scientific Research,1968      **Catalogue** Ann Arbor High School (Mich.),1903      **Q.M.C. Historical Studies ...** ,1944      The Development of Meat, Dairy, Poultry, and Fish Products for the Army Elliott Cassidy,1944      Cooperative Chemistry Lab Manual Cooper,2005-02 The laboratory course described in the lab manual emphasizes experimental design data analysis and problem solving Inherent in the design is the emphasis on communication skills both written and oral Students work in groups on open ended projects in which they are given an initial scenario and then asked to investigate a problem There are no formalized instructions and students must plan and carry out their own investigations      *A Companion To Interdisciplinary Stem Project-Based Learning* Mary Margaret Capraro,Robert Capraro,2016-07-18 This text contains 25 Project Based Learning PBL lessons written by a combination of undergraduate preservice teachers inservice teachers and graduate students Everyone who wrote a chapter strives to improve STEM education to help others implement standards based STEM instruction that takes learning in isolation to greater accountability through integrated and meaningful tasks that answer the question every teacher dreads When am I going to use this The PBLs were written to implement in middle and high school classrooms All of them are interdisciplinary in nature We have divided them into six themes construction and design water environment mixtures technology nutrition and genetics Each lesson contains a schedule at a glance and the well defined outcome so you can quickly see how a particular PBL fits into your curriculum Objectives are listed along with STEM connections written as objectives We have included all materials needed and then each day of activities including an imbedded engagement exploration explanation evaluation including rubrics and extension We have tried to include everything necessary for successful implementation This practical book is the perfect companion to the handbook for learning about implementing PBLs Project Based Learning An Integrated Science Technology Engineering and Mathematics

STEM Approach second edition      **Handbook of Formative Assessment** Heidi Andrade, Gregory J. Cizek, 2010-04-02

Formative assessment has recently become a focus of renewed research as state and federal policy makers realize that summative assessments have reached a point of diminishing returns as a tool for increasing student achievement. Consequently supporters of large scale testing programs are now beginning to consider the potential of formative assessments to improve student achievement. The mission of this handbook is to comprehensively profile this burgeoning field of study. Written by leading international scholars and practitioners each chapter includes a discussion of key issues that dominate formative assessment policy and practice today as well as those that are likely to affect research and practice in the coming years. Key features include:

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Illinois Chemistry Teacher , 1998      *The Science Teacher* , 1999

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