#### Secretary of

# Grade 6

Quiz 1 based on: 3.1.1 Particles in Motion

Tramentor 1, Work 2,

	Which of the following is not a common state of metapt?			
	18.0	Solid	46.7	Con
	36	Loped	#	Superfluid
	How do the particles of a gas differ from the particles of a biquid?			
	100	They are farther apart.	40	They are closer together.
	- 0-	They are smaller		They are larger
	If the particles of two substances have the same average kinetic energy, then			
	<ul> <li>a. the specific heats of the substances are identical.</li> </ul>			
	<ol> <li>the particles will not react chemically.</li> </ol>			
	<ul> <li>ii) the solvetances have equal thermal energies.</li> </ul>			
	2982	the substances have equal temperatures.		
		THE PERSON NAMED IN COLUMN TWO IS NOT THE OWNER.		

- 4. Why are the particles in matter always messing?
  - a. because they have energy
  - b. because they are being pushed

- c. because of their mans
- d. because of gravity
- 5. The temperature of a substance increases as ---
  - a. the substance expands.
  - b. the average kinetic energy of its particles increases.
  - 4. the substance's volume increases.
  - d. the substance's mass secreases.

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evolution synthetic biology and nanotechnology In many applications phage serves as a monodisperse long aspect nano scaffold of well defined shape Chemical or chenetic modifications of this scaffold are used to introduce the necessary functionalities such as fluorescent labels ligands that target specific proteins or peptides that promote formation of inorganic or organic nanostructures. We anticipate that the future holds development of new strategies for particle assembly site specific multi functional modifications and improvement of existing modification strategies. These improvements will render the production of filamentous phage templated materials safe and affordable allowing their applications outside of the laboratory.

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