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### CHEMISTRY

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	(A)	20-31	and the pk1 of the following samples by using pH paper/unitoresal indicator:	
		100	Dilute hydrochloric acid (BCI)	
		0.0	Dilute sedium kydrovide (NaOH) solutim	
		10110	Dilute ethanoic acid (CH <sub>2</sub> COOH) solution	
		(0v)	Lemon Juice	
		((4))	Winter	
		$\{(w_i)\}$	Dilute hydrogen carbonate solution.	
	(89)	Street	lying the properties of acids and bases (EEC) and NaOED on the basis of their reaction with	2
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		(c)	Solid sollium carbonate.	
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Yan Bai

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