



- 3.5 Substance exists as both a solid and a liquid ✓✓ or... is melting (2)
 3.6 37 °C ✓ (2)
 3.7 80 °C ✓ (1)
 3.8 An indication of the average kinetic energy ✓ of the particles in a substance. ✓ (2)
 3.9 Physical ✓ (1)
 3.10 In the solid state the intermolecular forces are the strongest and the particles cannot move. The heater provides thermal energy to the system, which is translated into kinetic energy (increase in temperature) until the intermolecular forces between the particles can be broken. ✓✓ At this point the thermal energy is transformed into latent energy to break the bonds, so no increase in temperature occurs. ✓✓ (4)

[23]

QUESTION 4

- 4.1
 4.1.1 Homogeneous ✓ (1)
 4.1.2 Homogeneous ✓ (1)
 4.1.3 Pure substance ✓ (1)
 4.2
 4.2.1 Potassium bromide ✓ (2)
 4.2.2 Calcium carbonate ✓ (2)
 4.2.3 Ammonium chloride ✓ (2)
 4.3
 4.3.1 ZnSO_4 ✓✓ (2)
 4.3.2 KMnO_4 ✓✓ (2)
 4.3.3 H_2O ✓✓ (2)

[17]

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web 1 carefully spread your sk atoms onto your paper towel count them you may separate the atoms into groups if it makes the counting faster 2 record the starting number of sk atoms in the skittlium atoms remaining row under the number zero on your data table 3 return all of the sk atoms to your cup 4

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web in decay a radioactive parent nucleus close nucleus the central part of an atom it contains protons and neutrons and has most of the mass of the atom the plural of nucleus is nuclei randomly