

CHEMISTRY PAPER 1 (THEORY)

Maximum Marks: 70

Time Allowed: Three hours

*Candidates are allowed additional 15 minutes for only reading the paper.
They must NOT start writing during this time.*

This paper is divided into four sections – A, B, C and D.

Answer all questions.

Section A consists of one question having sub-parts of one mark each.

Section B consists of ten questions of two marks each.

Section C consists of seven questions of three marks each, and

Section D consists of three questions of five marks each.

Internal choices have been provided in one question each in Section B,

Section C and Section D.

All working, including rough work, should be done on the same sheet as, and adjacent to the rest of the answer.

The intended marks for questions or parts of questions are given in brackets [].

Balanced equations must be given wherever possible and diagrams where they are helpful.

When solving numerical problems, all essential working must be shown.

In working out problems, use the following data:

Gas constant $R = 1.987 \text{ cal deg}^{-1} \text{ mol}^{-1} = 8.314 \text{ JK}^{-1} \text{ mol}^{-1}$

$= 0.0821 \text{ dm}^3 \text{ atm K}^{-1} \text{ mol}^{-1}$

$1 \text{ l atm} = 1 \text{ dm}^3 \text{ atm} = 101.3 \text{ J}$ $1 \text{ Faraday} = 96500 \text{ coulombs}$.

Avogadro's number $= 6.023 \times 10^{23}$.

SECTION A – 14 MARKS

Question 1

- (A) Fill in the blanks by choosing the appropriate word(s) from those given in the [4×1]
brackets:

[stable, low, aldehyde, unstable, 6, 4, ethane, Clemmensen's, 2, 3, carboxylic acid, high, propane, Rosenmund's]

- (i) The primary alcohols are easily oxidised first into _____ and then into _____.
- (ii) The intermediate activated complex in a chemical reaction is highly _____ due to _____ energy.
- (iii) The coordination number and oxidation state of the complex $\text{K}_4[\text{Fe}(\text{CN})_6]$ are _____ and _____ respectively.
- (iv) Propanone on reaction with zinc-amalgam in presence of conc. HCl gives _____ and the reaction is known as _____ reduction.

This Paper consists of 8 printed pages.

Grade 11 November 2019 Question Paper Chemistry

JE Gale



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Journal of Gas Lighting and Water Supply ,1892 *Catalogue of Civil Publications Relating to Agriculture, Forestry, Civic, Commerce, Finance, Legislation, Industry, Public Health, Railways, Science, Trade, Etc. Published by the Government of India* India. Government of India Publication Branch,1941 *Resources in education* ,1992-07 *The Louisiana Planter and Sugar Manufacturer* ,1888 **Nature** Sir Norman Lockyer,1873 **Research in Education** ,1974 Nuclear Science Abstracts ,1972 **The Education Gazette of the Province of the Cape of Good Hope** Cape of Good Hope (South Africa). Education Department,1980 Hoard's Dairyman ,1914 **The Journal of Gas Lighting, Water Supply & Sanitary Improvement** ,1892 Bulletin ,1926 **Industrial World** ,1898 **Record of Current Educational Publications ... Jan. 1912-Jan./Mar. 1932** United States. Office of Education,1921 Record of Current Educational Publications ,1921 **Catalogue of Government of India Civil Publications** India. Ministry of Works and Housing,1941

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