

FunBased Learning

Go to: Class Website → Used in Class → "Fun Based Learning"

Directions:

1. ...Go to Classic ChemBalancer
[when done with this activity, go to Review & Brain Boggle]
2. Click 'Directions'. Read and understand the directions.
3. Click 'OK', then on 'Start Game'
4. Try entering some numbers in the text boxes in front of each molecule.
What happens?
5. If you forget the directions, click on the 'How to Play the Game' link.
Click 'OK' when you finish reading them to return to the game.
6. When you think you have typed the right numbers in all the boxes,
click the 'Balanced' button.
7. If you didn't get it right, try again.
8. If you did get it right, then fill in the correct answers on this worksheet for #1.
9. Repeat steps 7-9 for the other 10 questions.
10. Now do the two problems on the back of this worksheet.

You can draw the molecules just like the program did to figure out the answer.
[11. When you're done, use the line in front of each problem to classify it:
S = synthesis, DR = double replacement, SR = single replacement,
A/R = add/base, D = decomposition, C = combustion]

Classic ChemBalancer Questions

Fill in the blanks below as you go through the game.
This is so I have a record that you did your assignment.

- "_____" 1. ____ Fe + ____ S → ____ FeS
- "_____" 2. ____ H₂ + ____ Cl₂ → ____ HCl
- "_____" 3. ____ Mg + ____ O₂ → ____ MgO
- "_____" 4. ____ O₂ + ____ H₂ → ____ H₂O
- "_____" 5. ____ H₂O → ____ H₂ + ____ O₂
- "_____" 6. ____ Ca + ____ H₂O → ____ Ca(OH)₂ + ____ H₂

- "_____" 7. ____ CH₄ + ____ O₂ → ____ CO₂ + ____ H₂O
- "_____" 8. ____ Na₂O₂ + ____ H₂SO₄ → ____ Na₂SO₄ + ____ H₂O₂
- "_____" 9. ____ N₂ + ____ H₂ → ____ NH₃
- "_____" 10. ____ Al + ____ O₂ → ____ Al₂O₃
- "_____" 11. ____ KMnO₄ → ____ K₂O + ____ MnO + ____ O₂

Draw the molecules just like the program did to figure out the answer to #1, and #11.

- "_____" 12. ____ Na + ____ H₂O → ____ NaOH + ____ H₂
- Fact for #12: Sodium metal, Na, is stored in kerosene so it won't react with water vapour. When added to water it reacts quickly to make hydrogen gas.

- "_____" 13. ____ H₂SO₄ + ____ NaOH → ____ Na₂SO₄ + ____ H₂O
- Fact for #13: This is an example of an acid base reaction. Acid + base → Salt + Water

Review ChemBalancer Questions

- "_____" 1. ____ S + ____ O₂ → ____ SO₂
- "_____" 2. ____ Na + ____ O₂ → ____ Na₂O₂
- "_____" 3. ____ H₂ + ____ O₂ → ____ H₂O
- "_____" 4. ____ Ag₂O → ____ Ag + ____ O₂
- "_____" 5. ____ Ba(OH)₂ + ____ H₃PO₄ → ____ BaHPO₄ + ____ H₂O
- "_____" 6. ____ NaOH + ____ H₃PO₄ → ____ Na₃HPO₄ + ____ H₂O

Fun Based Learning Review Chembalancer Answers

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Table of Contents Fun Based Learning Review Chembalancer Answers

1. Understanding the eBook Fun Based Learning Review Chembalancer Answers
 - The Rise of Digital Reading Fun Based Learning Review Chembalancer Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Fun Based Learning Review Chembalancer Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Fun Based Learning Review Chembalancer Answers
 - User-Friendly Interface

4. Exploring eBook Recommendations from Fun Based Learning Review Chembalancer Answers
 - Personalized Recommendations
 - Fun Based Learning Review Chembalancer Answers User Reviews and Ratings
 - Fun Based Learning Review Chembalancer Answers and Bestseller Lists
5. Accessing Fun Based Learning Review Chembalancer Answers Free and Paid eBooks
 - Fun Based Learning Review Chembalancer Answers Public Domain eBooks
 - Fun Based Learning Review Chembalancer Answers eBook Subscription Services
 - Fun Based Learning Review Chembalancer Answers Budget-Friendly Options
6. Navigating Fun Based Learning Review Chembalancer Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Fun Based Learning Review Chembalancer Answers Compatibility with Devices
 - Fun Based Learning Review Chembalancer Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Fun Based Learning Review Chembalancer Answers
 - Highlighting and Note-Taking Fun Based Learning Review Chembalancer Answers
 - Interactive Elements Fun Based Learning Review Chembalancer Answers
8. Staying Engaged with Fun Based Learning Review Chembalancer Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Fun Based Learning Review Chembalancer Answers
9. Balancing eBooks and Physical Books Fun Based Learning Review Chembalancer Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Fun Based Learning Review Chembalancer Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Fun Based Learning Review Chembalancer Answers
 - Setting Reading Goals Fun Based Learning Review Chembalancer Answers
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Fun Based Learning Review Chembalancer Answers
 - Fact-Checking eBook Content of Fun Based Learning Review Chembalancer Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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