

Melvin Smith

In Loving Memory of



Melvin Arnold Smith

Born
March 12, 1909
Mesa, Arizona

Passed
August 15, 2019
Mesa, Arizona

Funeral Service

Day of week, Month 16, 2019 at 11:00 AM
The Church of Jesus Christ of Latter-day Saints

Ward Name

Address

City, State

Presiding/Conducting Bishop (Name)

Family Prayer (Name), son

Organist (Name)

Chorister (Name)

Hymn #00 "Hymn Name"

Opening Prayer (Name), son-in-law

Life Story (Name), daughter

Funeral Home "Song Name"

Sung by (Name)

Accompanied by (Name)

(Name), son

Bishop (Name)

"Hymn Name"

Name, son-in-law

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tackles a very difficult topic Her direct and warm advice is a fabulous resource This book is a must read for anyone who will face the inevitability of honoring and celebrating the life of a deceased love one Peggy Post director of the Emily Post Institute and author of Emily Post s Etiquette 17th Edition Memorial Service ,1992 *Memorial Service Program* Frank Goad Clement,1969 Consists of the program for the memorial service for former Tennessee Governor Frank G Clement held at First United Methodist Church in Dickson Tennessee on November 6 1969 **Funeral Service Ticket and Memorial Service Program** William McKinley,Ida Saxton McKinley, Admission ticket to the funeral services of President McKinley in Canton Sept 19 1901 and memorial service program for Mrs Ida Saxton McKinley on June 2 1907

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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g}) \quad \Delta H = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta H = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{g}) \quad \Delta H = 283 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g})$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 2 \text{ H}_2\text{O}(\text{l}) \quad \Delta H = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{l}) \quad \Delta H = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + 1/2 \text{ O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta H = 285 \text{ kJ/mole}$

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