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DP Hallahan



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web the first two equations have been balanced for you alpha particle α 2 4 beta particle β gamma ray γ proton p neutron n 0

positron β^+ 0 1 30 1 65 1 2 63 2 15 0 30 0 65 30 9 29 2 2 12 6 15 32 30

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web identify the missing particle in the following nuclear reaction $^1_0\text{n} + ^{235}_{92}\text{U} \rightarrow ^{137}_{52}\text{Te} + ^{97}_{38}\text{Sr} + ^{97}_{40}\text{Zr}$

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web answer 4 correct in the reactants and products must be the same part b what is the value of in the following nuclear reaction $^{65}_{29}\text{Cu} + ^1_0\text{n} \rightarrow ^{65}_{28}\text{Ni} + ^1_0\text{n}$ express your answer as an integer hint 1 how to approach the problem use rule 2 the total of the subscripts atomic numbers mathematically this can be written as answer

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