

AS-Chapter 3 Handout

1. (a) Sodium reacts with cold water.

(b) What would you see as the reaction proceeds?

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CU

(c) Write the balanced chemical equation for this reaction.

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(d) Calculate the volume of gas produced if 3.0 g of sodium reacts with an excess of water.

(One mole of any gas at the temperature and pressure of the experiment occupies 24 dm^3 .)

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Total 7 marks

2. (a) (i) What is meant by the mass number of an atom?

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June 2013 Chemistr Unit 3b Paper 1a

Rachel Sandford



June 2013 Chemistr Unit 3b Paper 1al:

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