

Exercise 1: Density – Preparation of Solutions, KNO₃ and KCl, 2018

Provide data relevant to the following and calculate the mass ratio

Any substance having more density or heavier molecules is called solvent depending on the lighter molecules. Solute and solvent are combined together to form a homogeneous and transparent mixture. The process of combining the solute and solvent is called solution. The process of separating the solute and solvent is called separation.

1.1. Preparation of solutions

S. No.	Substance	Volume of water (ml)	Mass of substance (g)
1	KNO ₃		
2	KCl		
3	Water		
4	Total		

1.2. Preparation of solutions of potassium nitrate and potassium chloride

1.2.1. Preparation of potassium nitrate solution

1.2.1.1. Preparation of potassium nitrate solution

1.2.1.2. Preparation of potassium chloride solution

1.2.1.3. Preparation of potassium nitrate and potassium chloride solution

1.2.1.4. Preparation of potassium nitrate and potassium chloride solution

1.2.1.5. Preparation of potassium nitrate and potassium chloride solution

1.2.1.6. Preparation of potassium nitrate and potassium chloride solution

1.2.1.7. Preparation of potassium nitrate and potassium chloride solution

1.2.1.8. Preparation of potassium nitrate and potassium chloride solution

1.2.1.9. Preparation of potassium nitrate and potassium chloride solution



1.2.1.10. Preparation of potassium nitrate and potassium chloride solution

1.2.1.11. Preparation of potassium nitrate and potassium chloride solution

1.2.1.12. Preparation of potassium nitrate and potassium chloride solution

1.2.1.13. Preparation of potassium nitrate and potassium chloride solution

1.2.1.14. Preparation of potassium nitrate and potassium chloride solution

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