Honors Chemistry

Electrochemistry Problems

Name	

Date Period

The voltnic cell shows above is composed of the metals Nickel (Ni) and Calcions (Ca) and their nimate salts. Assume both metals are 2" in their ionic state. Pay attention to the direction of electronflow. Answer the following questions using the letters indicated above. The same letter may be used more than once.

- Soln bridge
- Nickel electrode
- 3. Calcium nitrate solution
- Calcium electrode
- Nickel nitrate solution
- Anode:
- Cineits order.
- 8. Electrode where oxidation occurs
- Electrode where reduction occurs
- 11. Electrode that gets smaller with time
- 12. Solution that receives positive ions from the salt bridge



ID.

«-Electron Flow Mictor, Lightbulb.

wolltmeter, etc.

A.

Write a condensed cell diagram for this reaction.

Write the oxidation half reaction for the cell.

Write the reduction half reaction for the cell.

For the following five questions identify the half reactions as either oxidized or reduced

16. Cl₂ + 2e²
$$\rightarrow$$
 2Cl²

which species is the redocing agent? Explain.

High School Electrochemistry Questions

M Mark

 ${\bf High\ School\ Electrochemistry\ Questions:}$

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