

QUESTION 2

	Test	Observation	Inference
	(a)(i) C + Conc. H_2SO_4 + strong heat	A copious brown irritating fume which turned moist blue litmus paper to red	Gas is NO_2 NO_3^- present
	(b)(i) C + water + NaOH in drops, in excess	white gelatinous precipitate precipitate dissolved	Al^{3+} or Zn^{2+}
	(ii) C + water + NH_3 in drops, in excess	white gelatinous precipitate precipitate dissolved	Al^{3+} or Zn^{2+}
Trickle slowly down	(iii) C + water + freshly prepared $FeSO_4$ + Conc. H_2SO_4	A brown ring is formed at the junction of the acid and aqueous solution.	NO_3^-
	(c)(i) D + water	D dissolved	soluble
	(ii) Solution D + dil HCl	Colourless gas, smells like rotten egg, turns moist blue litmus paper to red and lead (II) ethanoate paper to black	H_2S present
	(iii) Solution D + NaOH in drops, in excess	white precipitate precipitate insoluble	Ca^{2+} or Pb^{2+}
	(iv) solution D + $(NH_4)_2CO_3$	white precipitate	Ca^{2+}

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