

General Instructions:

Read the following instructions very carefully and follow them:

- This question paper contains 35 questions. All questions are compulsory.
- Question paper is divided into FIVE sections Section A, B, C, D and E.
- (iii) In Section A: Question number 1 to 18 are Multiple Choice (MCQ) type questions carrying 1 mark each.
- (iv) In Section B: Question number 19 to 25 are Short Answer-1 (SA-1) type questions carrying 2 marks each.
- (v) In Section C: Question number 26 to 30 are Short Answer-2 (SA-2) type questions carrying 3 marks each.
- (vi) In Section D: Question number 31 to 33 are Long Answer (LA) type questions carrying 5 marks each.
- (vii) In Section E: Question number 34 and 35 are Case-Based questions carrying 4 marks each.
- (viii) There is no overall choice. However, an internal choice has been provided in 2 questions in Section—B, 2 questions in Section—C, 3 questions in Section—D and 2 questions in Section—E.
- (ix) Use of calculators is NOT allowed.

$$c = 3 \times 10^8 \text{ m/s}$$

 $h = 6.63 \times 10^{-34} \text{ Js}$
 $e = 1.6 \times 10^{-19} \text{ C}$

$$\mu_0 = 4\pi \times 10^{-7} \text{ T m A}^{-1}$$

$$\epsilon_0 = 8.854 \times 10^{-12}~\mathrm{C^2~N^{-1}~m^{-2}}$$

$$\frac{1}{4\pi\epsilon_0}$$
 = 9 × 10⁹ N m² C⁻²

Mass of electron (m_e) = 9.1×10^{-31} kg

Mass of neutron =
$$1.675 \times 10^{-27}$$
 kg

Mass of proton =
$$1.673 \times 10^{-27}$$
 kg

Avogadro's number =
$$6.023 \times 10^{23}$$
 per gram mole

Boltzmann constant =
$$1.38 \times 10^{-23} \, \mathrm{JK^{-1}}$$

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Table of Contents Matric 2013 June Physics Question Paper

- 1. Understanding the eBook Matric 2013 June Physics Question Paper
 - The Rise of Digital Reading Matric 2013 June Physics Question Paper
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Matric 2013 June Physics Question Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Matric 2013 June Physics Question Paper
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Matric 2013 June Physics Question Paper
 - Personalized Recommendations
 - Matric 2013 June Physics Question Paper User Reviews and Ratings
 - Matric 2013 June Physics Question Paper and Bestseller Lists
- 5. Accessing Matric 2013 June Physics Question Paper Free and Paid eBooks
 - Matric 2013 June Physics Question Paper Public Domain eBooks
 - Matric 2013 June Physics Question Paper eBook Subscription Services
 - Matric 2013 June Physics Question Paper Budget-Friendly Options
- 6. Navigating Matric 2013 June Physics Question Paper eBook Formats

- ePub, PDF, MOBI, and More
- Matric 2013 June Physics Question Paper Compatibility with Devices
- Matric 2013 June Physics Question Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Matric 2013 June Physics Question Paper
 - Highlighting and Note-Taking Matric 2013 June Physics Question Paper
 - Interactive Elements Matric 2013 June Physics Question Paper
- 8. Staying Engaged with Matric 2013 June Physics Question Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - $\circ\,$ Following Authors and Publishers Matric 2013 June Physics Question Paper
- 9. Balancing eBooks and Physical Books Matric 2013 June Physics Question Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Matric 2013 June Physics Question Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Matric 2013 June Physics Question Paper
 - Setting Reading Goals Matric 2013 June Physics Question Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Matric 2013 June Physics Question Paper
 - Fact-Checking eBook Content of Matric 2013 June Physics Question Paper
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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