

Paper 1

Questions are applicable for both core and extended candidates

- 1 Which statement describes methane?
- A It is an alcohol.
 - B It is an unsaturated molecule.
 - C It contains carbon, hydrogen and oxygen atoms only.
 - D Its molecules contain four single covalent bonds.
- 2 Which equation represents the cracking of an alkane?
- A $3\text{C}_2\text{H}_4 \rightarrow \text{C}_6\text{H}_{12}$
 - B $\text{C}_6\text{H}_{12} + \text{H}_2 \rightarrow \text{C}_6\text{H}_{14}$
 - C $\text{C}_6\text{H}_{14} \rightarrow 6\text{C} + 7\text{H}_2$
 - D $\text{C}_6\text{H}_{14} \rightarrow \text{C}_2\text{H}_4 + \text{C}_4\text{H}_{10}$
- 3 Which statement about ethane is correct?
- A It decolourises bromine water.
 - B It burns in excess oxygen to form water and carbon dioxide.
 - C Its molecular formula is C_2H_4 .
 - D Its atoms are joined together by ionic bonding.
- 4 Some properties of colourless liquid L are listed.
- It boils at 65°C .
 - When added to water, two layers form which do not mix.
 - It does not react with sodium carbonate.
 - It has no effect on bromine water.

What is L?

- A ethanol
- B hexane
- C hexene
- D ethanoic acid

Igcse Chemistry Examination Paper Jan 24

Bryan Earl, Doug Wilford



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