

Name: _____ Date: _____ Period: _____

NOVA: Hunting the Elements

Part 1: Basic Chemistry

Blank Periodic Table

Fill in the parts of this periodic table covered by the documentary.

1 H 1.00794																	2 He 4.002602	
3 Li 6.941	4 Be 9.00918											5 B 10.811	6 C 12.011	7 N 14.00643	8 O 15.999	9 F 18.998403	10 Ne 20.1797	
11 Na 22.98976928	12 Mg 24.304											13 Al 26.9815386	14 Si 28.08558	15 P 30.973762	16 S 32.06	17 Cl 35.453	18 Ar 39.948	
19 K 39.0983	20 Ca 40.078	21 Sc 44.955912	22 Ti 47.88	23 V 50.9415	24 Cr 51.9961	25 Mn 54.938045	26 Fe 55.845	27 Co 58.933195	28 Ni 58.6934	29 Cu 63.546	30 Zn 65.38	31 Ga 69.723	32 Ge 72.64	33 As 74.9216	34 Se 78.96	35 Br 79.904	36 Kr 83.80	
37 Rb 85.4678	38 Sr 87.62	39 Y 88.90584	40 Zr 91.224	41 Nb 92.90638	42 Mo 95.94	43 Tc 98.9062	44 Ru 101.07	45 Rh 102.9055	46 Pd 106.90558	47 Ag 107.8642	48 Cd 112.411	49 In 114.818	50 Sn 118.710	51 Sb 121.757	52 Te 127.6	53 I 126.905	54 Xe 131.29	
55 Cs 132.90545196	56 Ba 137.327	57-70 (*)	71 Lu 174.967	72 Hf 178.49	73 Ta 180.94788	74 W 183.84	75 Re 186.207	76 Os 190.23	77 Ir 192.222	78 Pt 195.084	79 Au 196.966569	80 Hg 200.59	81 Tl 204.3833	82 Pb 207.2	83 Bi 208.9804	84 Po 209	85 At 210	86 Rn 222
87 Fr 223	88 Ra 226	89-102 (*) (*)	103 Lr 260	104 Rf 261	105 Db 262	106 Sg 266	107 Bh 264	108 Hs 277	109 Mt 268	110 Uun 288	111 Uuu 289	112 Uub 285	114 Uuq 289					

^a Lanthanide series.

57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb
89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No

[illegible]

Gold - Au

1. Write the number of subatomic particles in gold:
 - a. Protons;
 - b. Neutrons;
 - c. Electrons;
2. Give one property of gold.
3. How much gold is in one ton of the mined rock?

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Table of Contents Hunting The Elements Worksheet Answer Key

1. Understanding the eBook Hunting The Elements Worksheet Answer Key
 - The Rise of Digital Reading Hunting The Elements Worksheet Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Hunting The Elements Worksheet Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Hunting The Elements Worksheet Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Hunting The Elements Worksheet Answer Key
 - Personalized Recommendations
 - Hunting The Elements Worksheet Answer Key User Reviews and Ratings

- Hunting The Elements Worksheet Answer Key and Bestseller Lists
- 5. Accessing Hunting The Elements Worksheet Answer Key Free and Paid eBooks
 - Hunting The Elements Worksheet Answer Key Public Domain eBooks
 - Hunting The Elements Worksheet Answer Key eBook Subscription Services
 - Hunting The Elements Worksheet Answer Key Budget-Friendly Options
- 6. Navigating Hunting The Elements Worksheet Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Hunting The Elements Worksheet Answer Key Compatibility with Devices
 - Hunting The Elements Worksheet Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Hunting The Elements Worksheet Answer Key
 - Highlighting and Note-Taking Hunting The Elements Worksheet Answer Key
 - Interactive Elements Hunting The Elements Worksheet Answer Key
- 8. Staying Engaged with Hunting The Elements Worksheet Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Hunting The Elements Worksheet Answer Key
- 9. Balancing eBooks and Physical Books Hunting The Elements Worksheet Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Hunting The Elements Worksheet Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Hunting The Elements Worksheet Answer Key
 - Setting Reading Goals Hunting The Elements Worksheet Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Hunting The Elements Worksheet Answer Key
 - Fact-Checking eBook Content of Hunting The Elements Worksheet Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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