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web answer the correct option is c explanation the molarity of a mixture m_{mix} can be calculated using the following formula $m_{\text{mix}} = \frac{m_1 + m_2 + m_3}{V_1 + V_2 + V_3}$ $1.5 \times 480 + 1.2 \times 520 + 480 + 520 + 720 + 624 = 1000$ 1.344 M q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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web molality differs from molarity only in the denominator while molarity is based on the liters of solution molality is based on the kilograms of solvent concentrations expressed in molality are used when studying properties of solutions related to vapor pressure and temperature changes

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