

A New Hydrogen Purification Process

COMMERCIAL DEVELOPMENT OF PALLADIUM ALLOY DIFFUSION CELLS

By J. B. Hunter, Ph.D.

J. Bishop and Co Platinum Works, Malvern, Pennsylvania

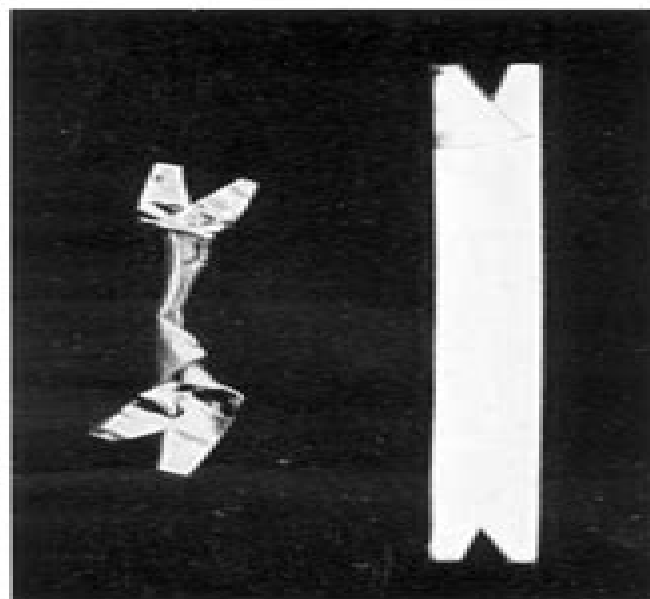
The discovery of a palladium alloy having remarkable stability as well as a hydrogen transfer rate more than twice that of pure palladium underlies the development of a new industrial process for the separation or purification of hydrogen.

The selective diffusion of hydrogen through a palladium membrane has been well known since it was first demonstrated by Thomas Graham in 1866; equally well known has been the dimensional change that occurs when palladium is alternately heated and cooled in hydrogen. This deformation, which is generally understood to be the result of an α - β phase change in the metal as the hydrogen is occluded and released, has severely limited the development of diffusion equipment for the commercial separation and purification of hydrogen.

Effect of Alloying Additions on Palladium

It had also been generally believed that the alloying of another metal with palladium would serve only to reduce its ability to transfer hydrogen, and investigations with additions of copper, gold, nickel, iron, platinum and ruthenium have in fact proved this to be the case. If, however, silver is alloyed with palladium some unexpected results are obtained. Some compositions show a reduced permeability, others give essentially the same permeability as the parent metal, but in a very narrow critical range the rate of hydrogen transfer is found to increase remarkably.

With the alloy finally established as the most effective, the hydrogen permeability at 500°C is about 25 per cent greater than for palladium, while at 300°C the difference in performance has increased to some 150 per cent. Although this remarkable increase in permeability of



Specimens of pure palladium (left) and of the silver-palladium alloy (right), initially the same size, after thirty cycles of heating and cooling in hydrogen

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