

Question Number	Acceptable Answers	Mark
24(b)(i)	EITHER $\begin{array}{rcl} \text{C} & : & \text{H} & : & \text{O} \\ \text{mol } 64.9 & : & 13.5 & : & 21.6 \\ & 12 & 1 & 16 \\ = & 5.408 & : & 13.5 & : & 1.35 \\ = & 4.006 & : & 10 & : & 1 \\ = & 4 & : & 10 & : & 1 \end{array}$ use of 74 to show molecular formula is $\text{C}_4\text{H}_{10}\text{O}$ eg M_r is $(4 \times 12) + (10 \times 1) + 16 = 74$ OR C atoms = $\frac{64.9 \times 74}{100 \times 12} = 4$ H atoms = $\frac{13.5 \times 74}{100 \times 1} = 10$ O atoms = $\frac{21.6 \times 74}{100 \times 16} = 1$ This may be done in 2 steps eg C $\frac{64.9 \times 74}{100} = 48$ $\frac{48}{12} = 4$ All 3 correct scores 2 Any 2 correct scores 1 OR % C = $\frac{48 \times 100}{74} = 64.9$ % H = $\frac{10 \times 100}{74} = 13.5$ % O = $\frac{16 \times 100}{74} = 21.6$ All 3 correct scores (2) Any 2 correct scores (1)	(2)

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Table of Contents January 2015 Ial C4 Edexcel

1. Understanding the eBook January 2015 Ial C4 Edexcel
 - The Rise of Digital Reading January 2015 Ial C4 Edexcel
 - Advantages of eBooks Over Traditional Books
2. Identifying January 2015 Ial C4 Edexcel
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an January 2015 Ial C4 Edexcel
 - User-Friendly Interface
4. Exploring eBook Recommendations from January 2015 Ial C4 Edexcel
 - Personalized Recommendations
 - January 2015 Ial C4 Edexcel User Reviews and Ratings
 - January 2015 Ial C4 Edexcel and Bestseller Lists
5. Accessing January 2015 Ial C4 Edexcel Free and Paid eBooks
 - January 2015 Ial C4 Edexcel Public Domain eBooks
 - January 2015 Ial C4 Edexcel eBook Subscription Services
 - January 2015 Ial C4 Edexcel Budget-Friendly Options
6. Navigating January 2015 Ial C4 Edexcel eBook Formats

- ePub, PDF, MOBI, and More
- January 2015 Ial C4 Edexcel Compatibility with Devices
- January 2015 Ial C4 Edexcel Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of January 2015 Ial C4 Edexcel
 - Highlighting and Note-Taking January 2015 Ial C4 Edexcel
 - Interactive Elements January 2015 Ial C4 Edexcel
- 8. Staying Engaged with January 2015 Ial C4 Edexcel
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers January 2015 Ial C4 Edexcel
- 9. Balancing eBooks and Physical Books January 2015 Ial C4 Edexcel
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection January 2015 Ial C4 Edexcel
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine January 2015 Ial C4 Edexcel
 - Setting Reading Goals January 2015 Ial C4 Edexcel
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of January 2015 Ial C4 Edexcel
 - Fact-Checking eBook Content of January 2015 Ial C4 Edexcel
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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