PART I: CHEMISTRY

SECTION - I

Single Correct Choice Type

This section contains 8 multiple choice questions. Each question has 4 choices (A), (B), (C) and (D) for its answer, out of which ONLY ONE is correct.

 Given that the abundances of isotopes ¹⁴Fe, ¹⁶Fe and ¹⁰Fe are 5%, 90% and 5%, respectively, the atomic mass of Fe is

(A) 55.85

(8) 55.95

(CD 55.75)

1839 545,000

Answer



The term that corrects for the attractive forces present in a real gas in the van der Waals equation is

CAD IND

GRO Sent

 $\frac{(C)}{V^2}$

(ID) - n.5-

Answer



 Among the electrolytes Na₂SO₄, CaCl₂, Al₂(SO₄)₁ and NH₄Cl, the most effective congulating agent for Sb₂S₃ set is

(A) Na₀SO₄

(B) CaCl₂

 $(C) = AI_2(SO_4)_3$

CIDO NULL CU

Answer



 The Henry's law constant for the solubility of N₂ gas in water at 298 K is 1.0×10° atm. The mole fraction of N₂ in air is 0.8. The number of moles of N₂ from air dissolved in 10 moles of water at 298 K and 5 atm pressure is

(A) 4.0 × 10⁻⁴

(B) 4.0 × 10 °

OCT 5.0 × 10 *

CDO 4.0 × 10 **

Answer

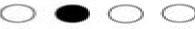


The reaction of P₄ with X leads selectively to P₄O_n. The X is

(A) Dry O₂ (C) Moint O₂ (B) A mixture of O. and N.

(D) O₂ in the presence of aqueous NaOH

Answer



E 45.3



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B Lingard

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